

New Concepts in Mass and Energy Transport using Graphene and Carbon Nanotubes

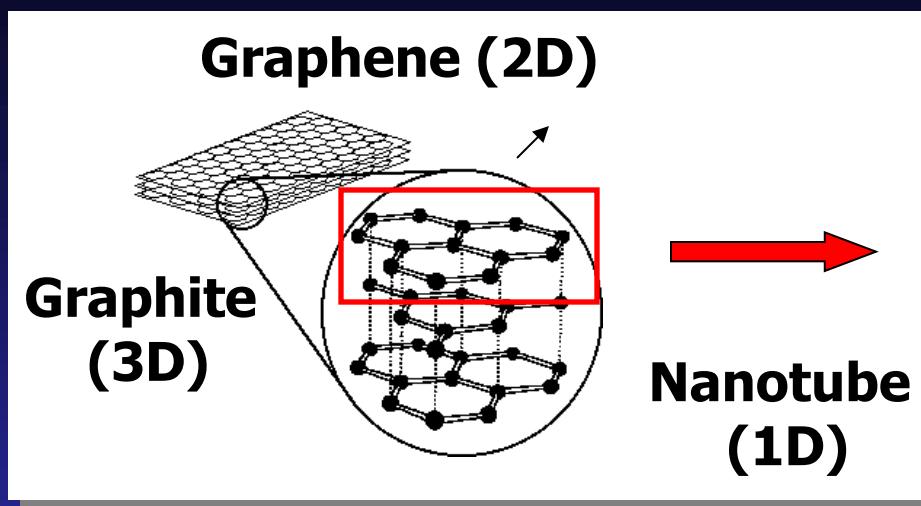


Michael S. Strano

**Department of Chemical Engineering
Massachusetts Institute of Technology**

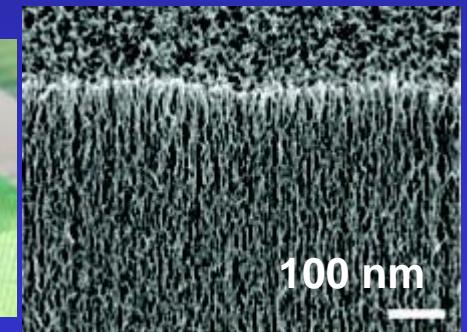
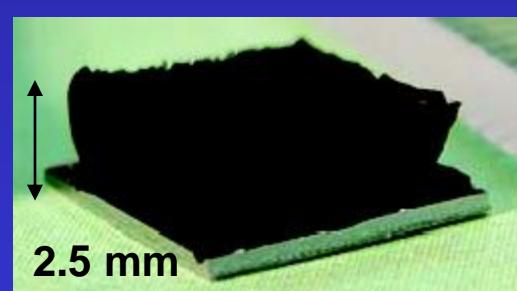
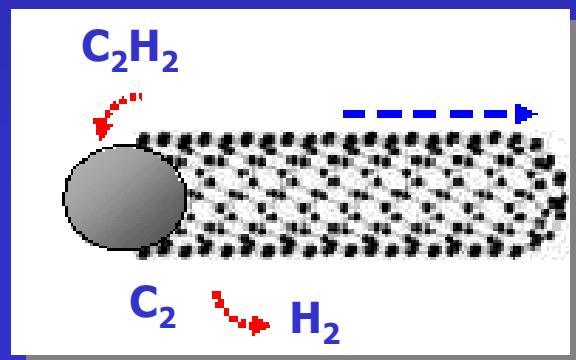
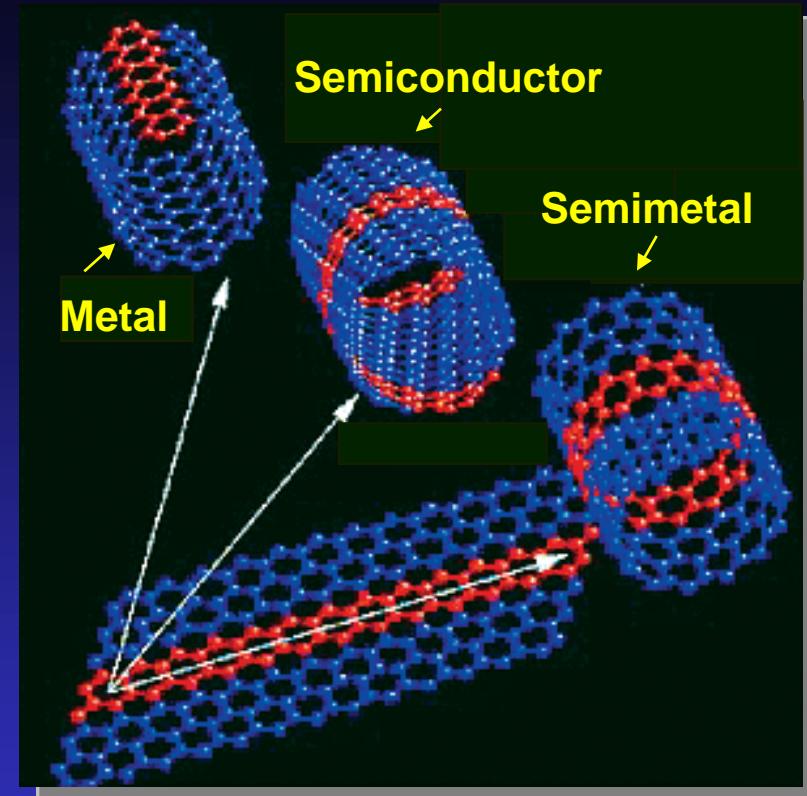
**W. Choi, C. Y. Lee, J. Abrahamson, N. Nair, C.
Song, J. H. Han, G. Paulus, A. Boghossian**

Carbon Nanotubes and Graphene



Electrons restricted to single dimension
Diameter ~1 nm

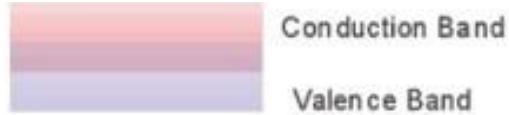
Synthesis (now > 1000 tons/year)
Stratis V. Sotirchos: electric arc synthesis
Over metal (Fe, Co, Mo) nanoparticle



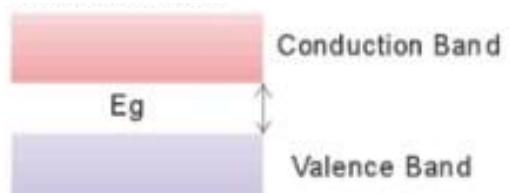
Length can range from 10 nm to 1 cm

Diversity of Electronic Structure

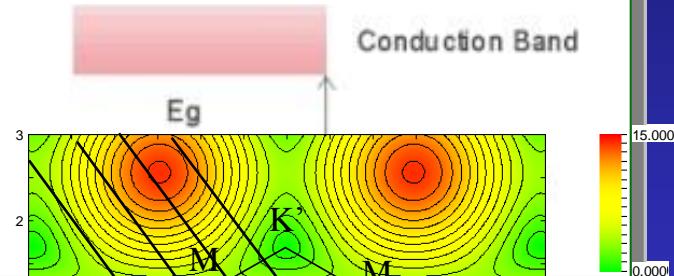
Metal



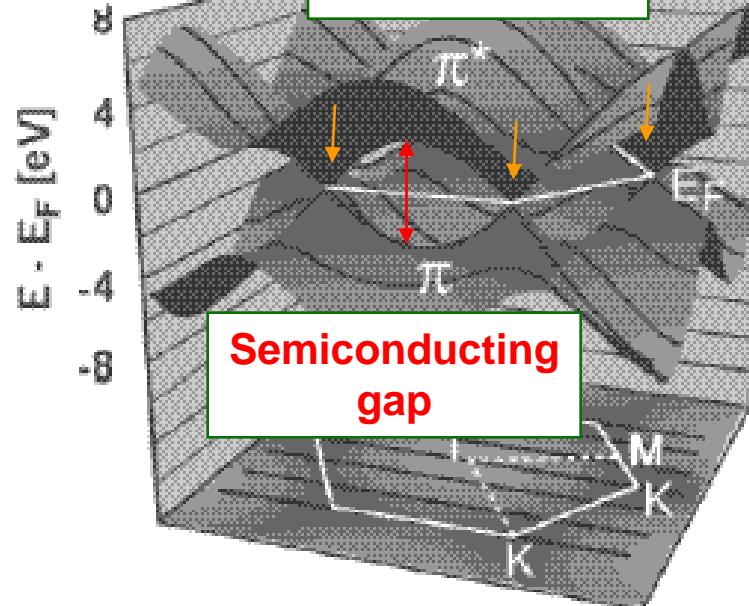
Semi-conductor



Insulator

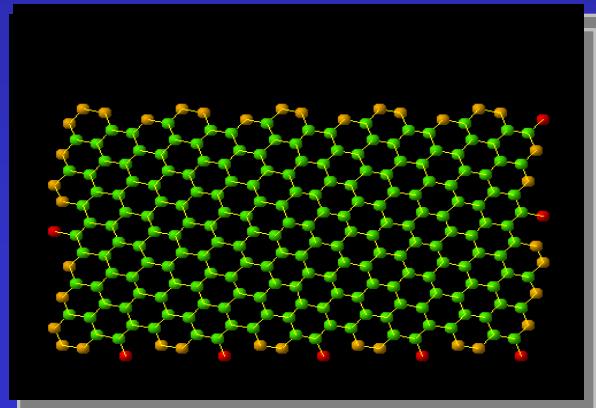


Metallic Nodes

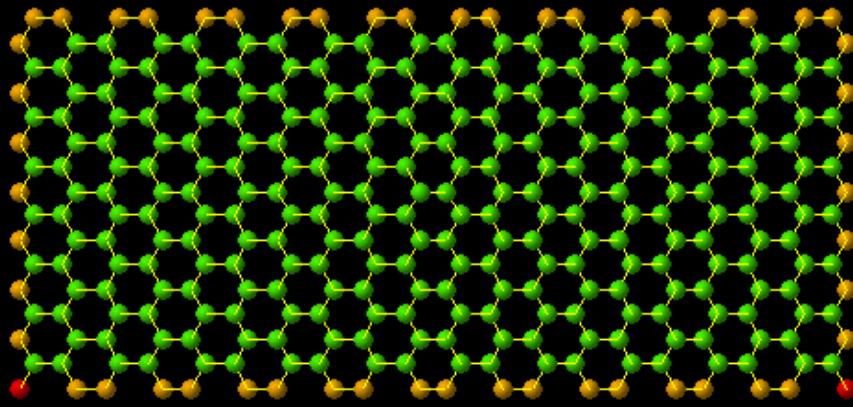


Rolling into cylinder
constrains electrons
to 1-D

Interstitial point
→ semimetallicizing

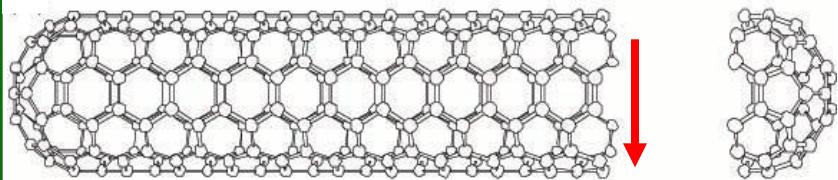


The Chirality Vector (n,m)



(10,10)

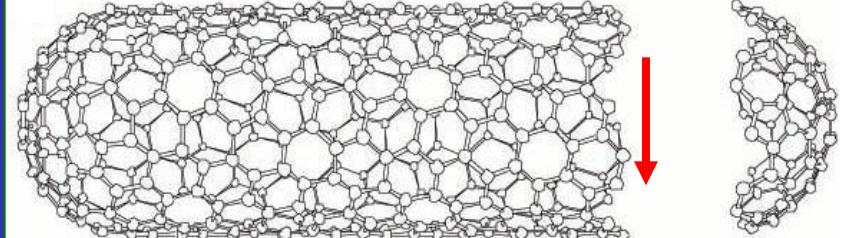
(n,m) where $n = m$



"armchair" nanotube, 1-D metal
if $(n - m)/3 = \text{integer}$ then metallic

Energy (eV)

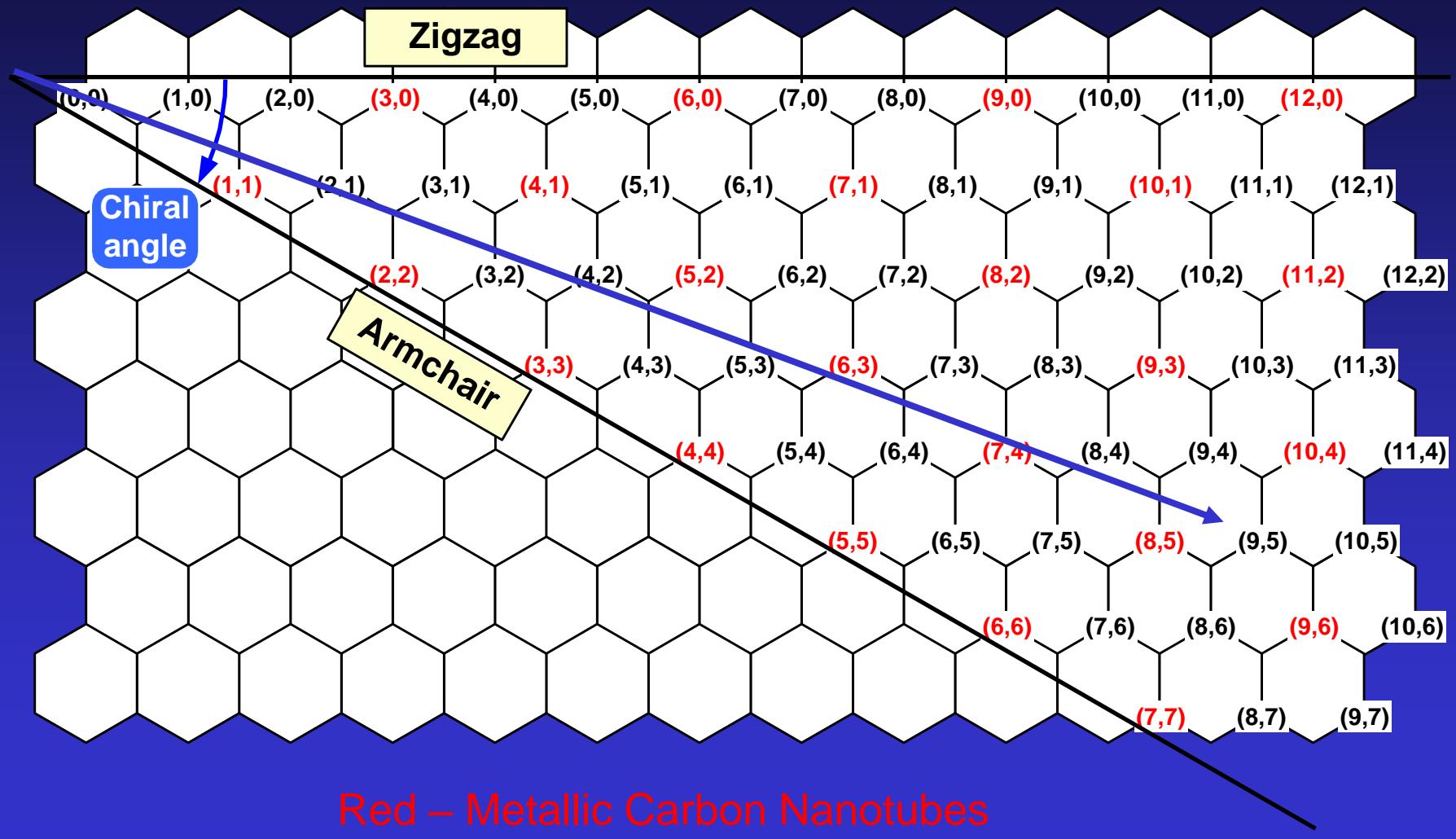
(10,5)

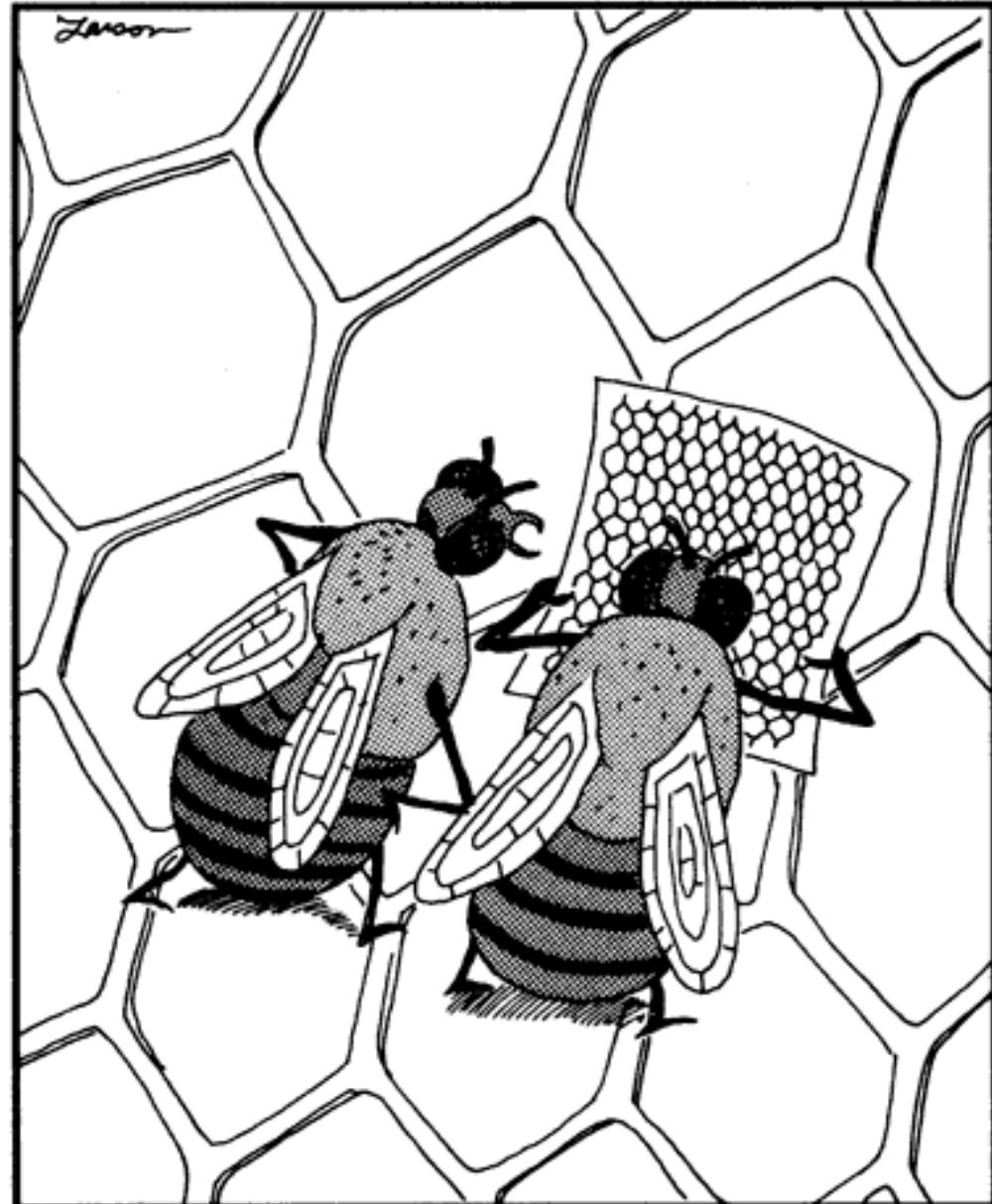


chiral nanotube, 1-D metal or
semiconductor if $(n - m)/3 \neq \text{integer}$

Energy (eV)

Construction of Nanotubes from a Graphene Sheet





"Face it, Fred—you're lost!"

Applications of Graphene and Carbon Nanotubes

Ideal Molecular Wires and Films

High current density 10^9 A/cm^2 or $30 \mu\text{A}$ per nanotube

High electron mobility $20,000 \text{ cm}^2/\text{V}\cdot\text{s}$

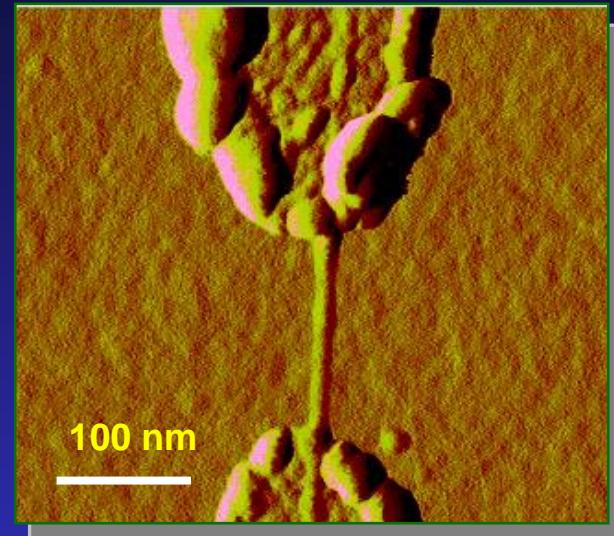
V. Perebeinos, J. Tersoff, and Ph. Avouris,
Phys. Rev. Lett. **94**, 086802 (2005).

Unique Field Effect Transistors

Graphene bilayers – tunable bands

Shih, C. J. and Strano, M.S. *Nanoletters* (2013) in press.

Shih, C. J. and Strano, M. S. *Langmuir*, 28, 22 (2012) 8579



Near Infrared Absorption/Emission

Tissue implantable SWNT glucose sensors

Barone, P. and Strano, M. S. *Nature Materials* (2005)

Yoon, H. and Strano, M. S. *Angewandte Chemie* 50, 8
(2011) 1828

Structural Applications

High elastic modulus 1 TPa

Highest strength to weight ratio

Salvetat, J. P. et. al. *Physical Review Letters*
1999, 82, 944-947.

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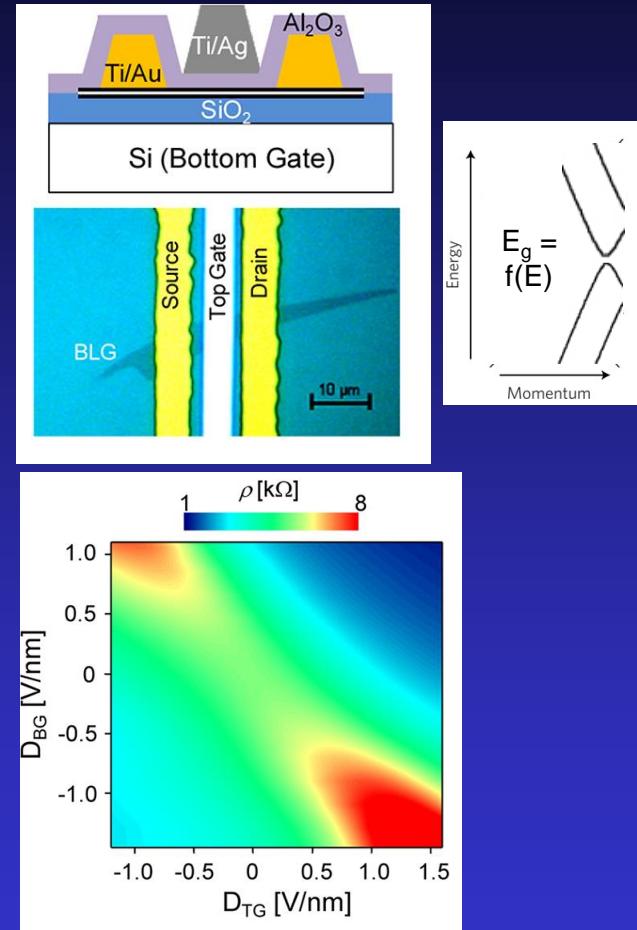
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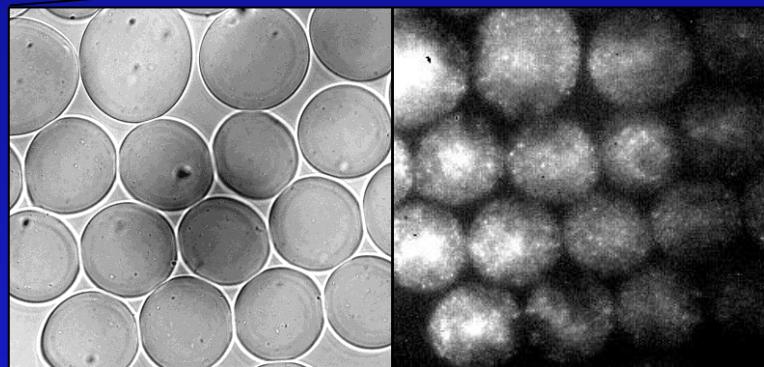
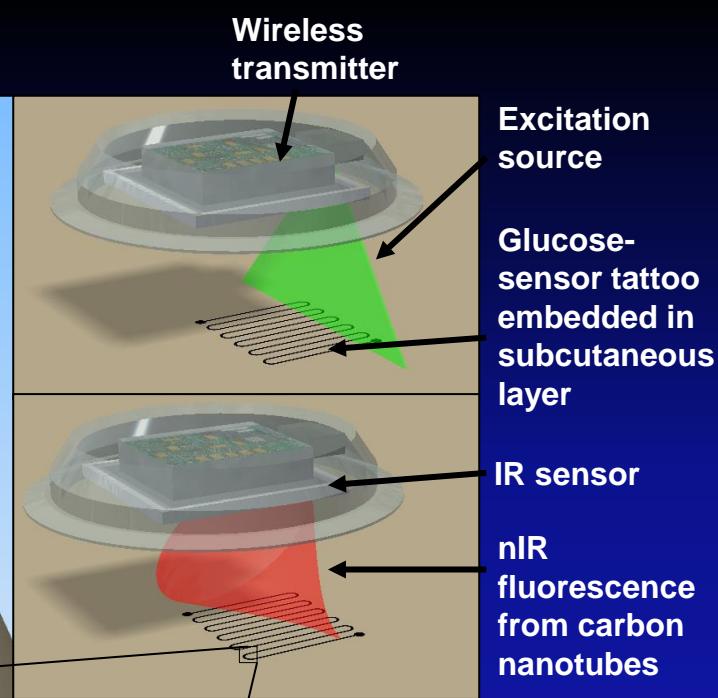
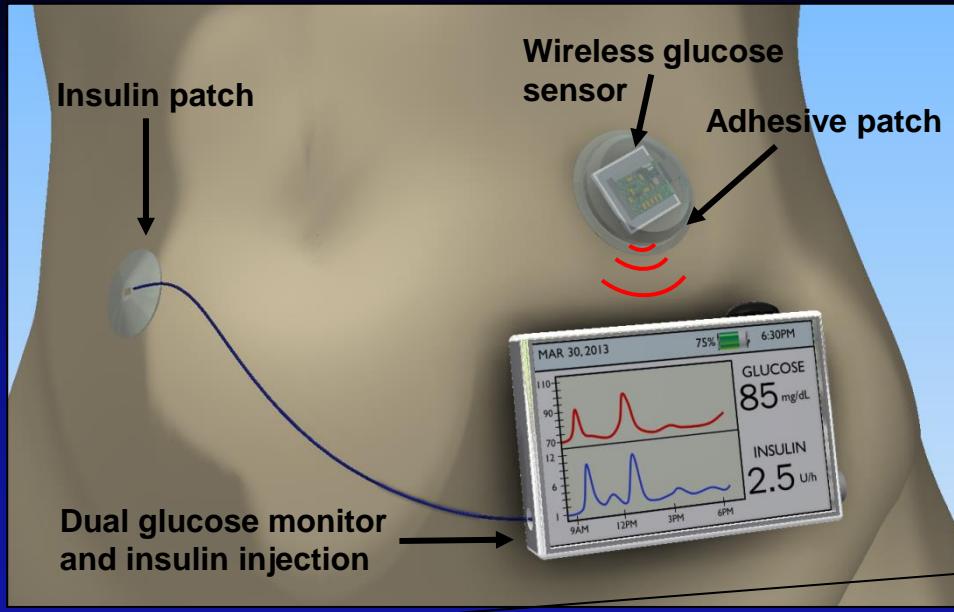
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Applications of Graphene and Carbon Nanotubes



Alginate
microparticles
with carbon
nanotubes

Optical

nIR Fluorescence

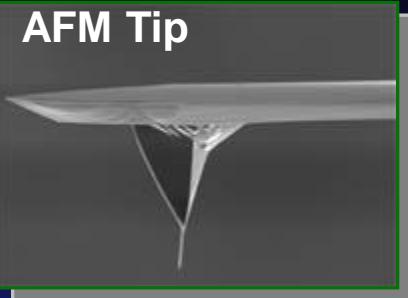
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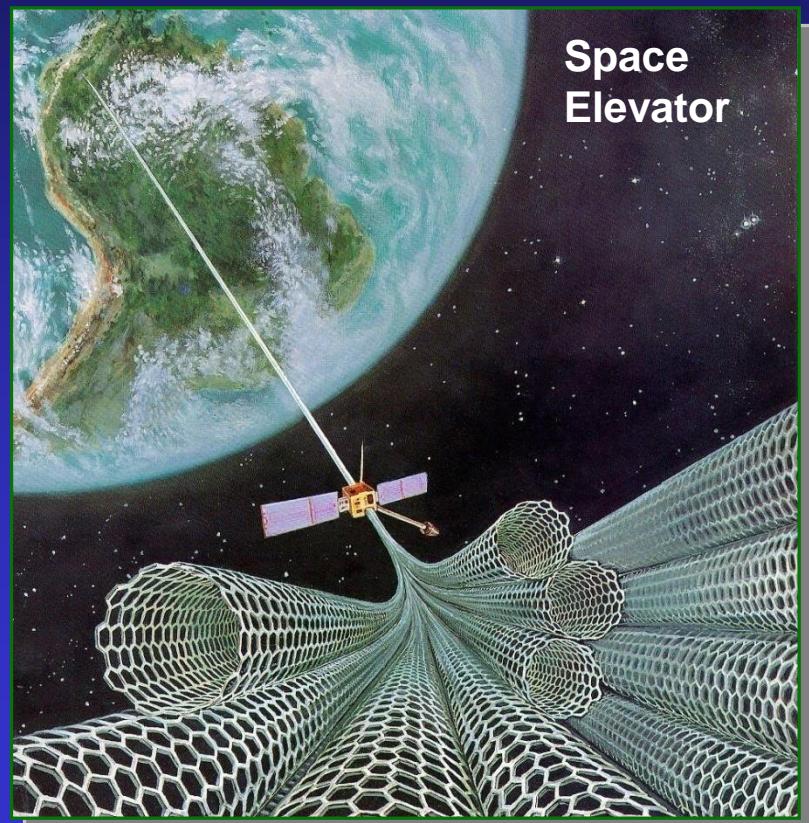


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New Concepts in Mass and Energy Transport using Graphene and Carbon Nanotubes

- Coherence resonance in **molecular transport** through a single walled carbon nanotube nanopore
- Energy storage and generation using **thermopower waves**
- Near infrared **fluorescent sensors** with single molecule sensitivity for studying biological signaling fluxes: reactive oxygen and nitric oxide signaling in Epidermal Growth Factor Receptor (EGFR)

New Concepts in Mass and Energy Transport within Carbon Nanotubes

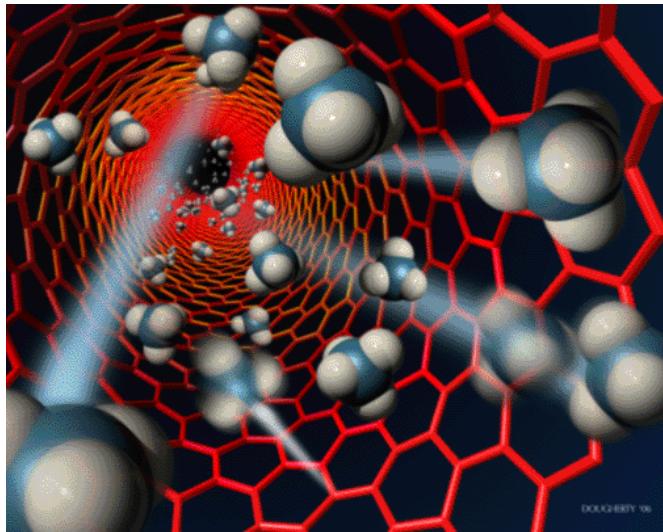
- Coherence resonance in **molecular transport** through a single walled carbon nanotube nanopore
- Energy storage and generation using thermopower waves
- Near infrared fluorescent sensors with single molecule sensitivity for studying biological signaling fluxes: reactive oxygen and nitric oxide signaling in Epidermal Growth Factor Receptor (EGFR)

Molecular Transport through a Single Nanotube

Substantial interest in understanding molecular transport within nanotubes

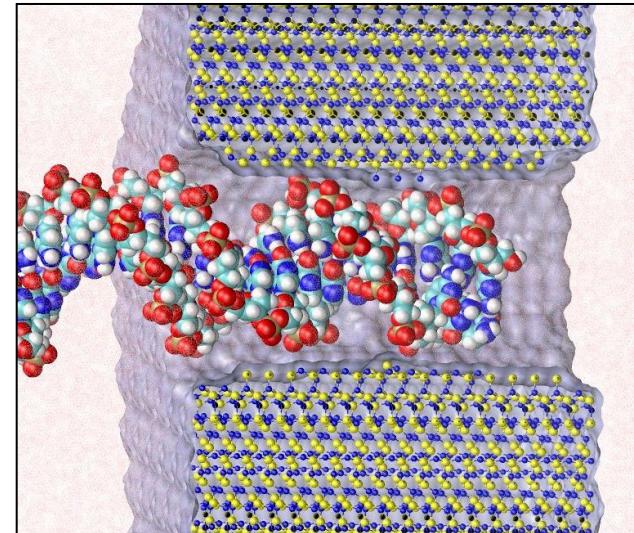
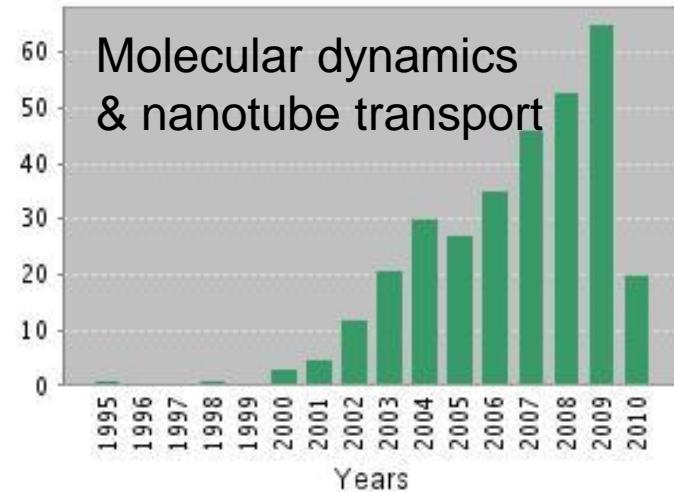
Most studies have been computational;
Many applications

- Fuel cells membranes
- Energy storage
- DNA sequencing
- Water desalination



scitizen.com/.../slippery-nanopipes_a-5-131.html

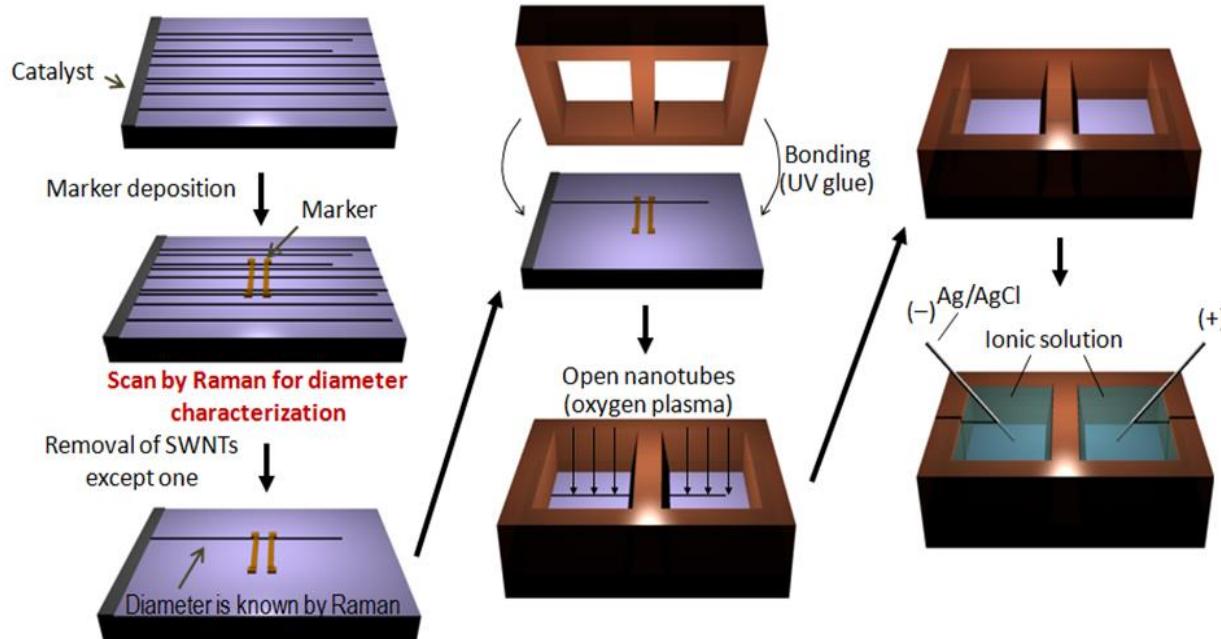
Published Items in Each Year



physics.illinois.edu/people/aksimentiev/

Creating and Characterizing SWNT Ion Channels

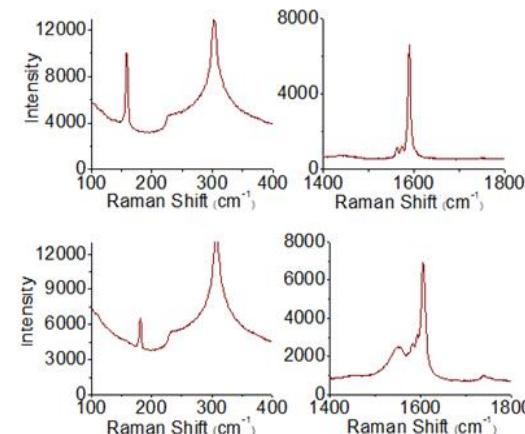
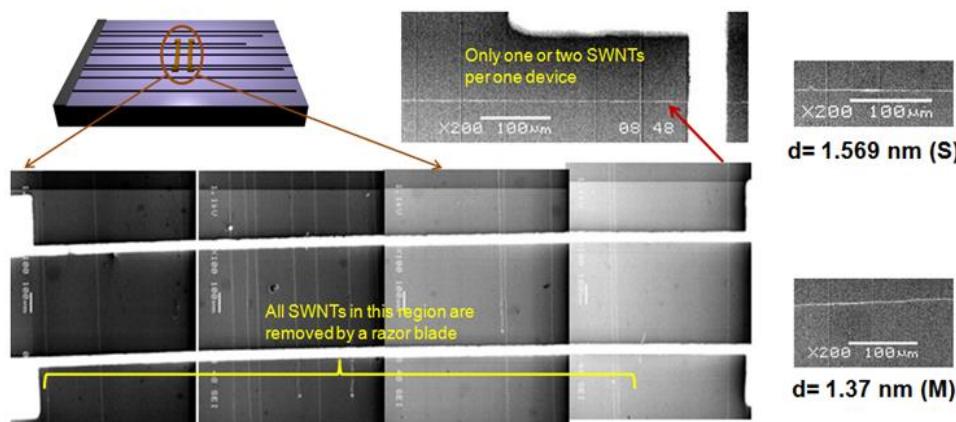
Lee, CY, Choi W, Han, JH, Strano MS: Science, 239, 1320 - 1324 (2010).
Choi, W., Shimizu, S., Strano, M.S. in preparation



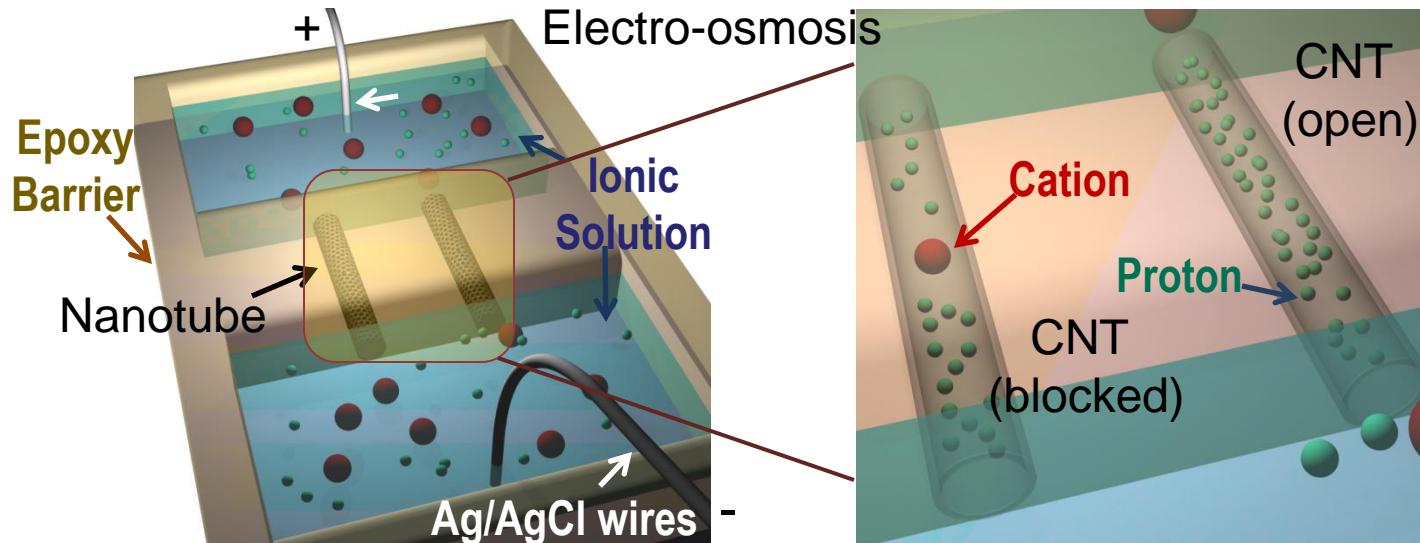
Won Joon
Choi



Chang
Young Lee



Ion Transport in the Nanotube Interior



2 sec



Pure Water (without electrolyte, 1000 mV)

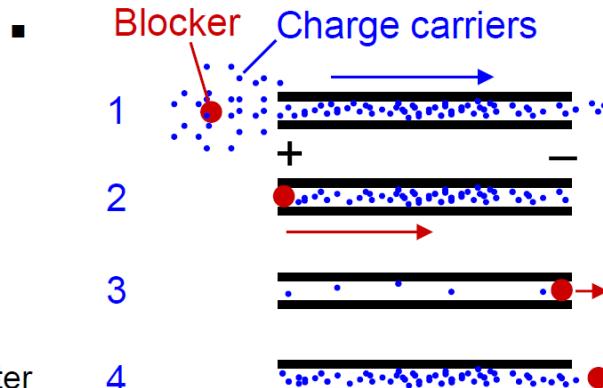
2 sec



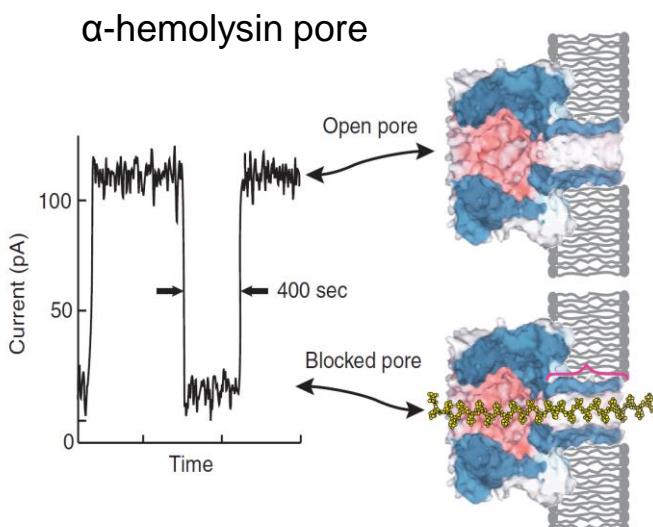
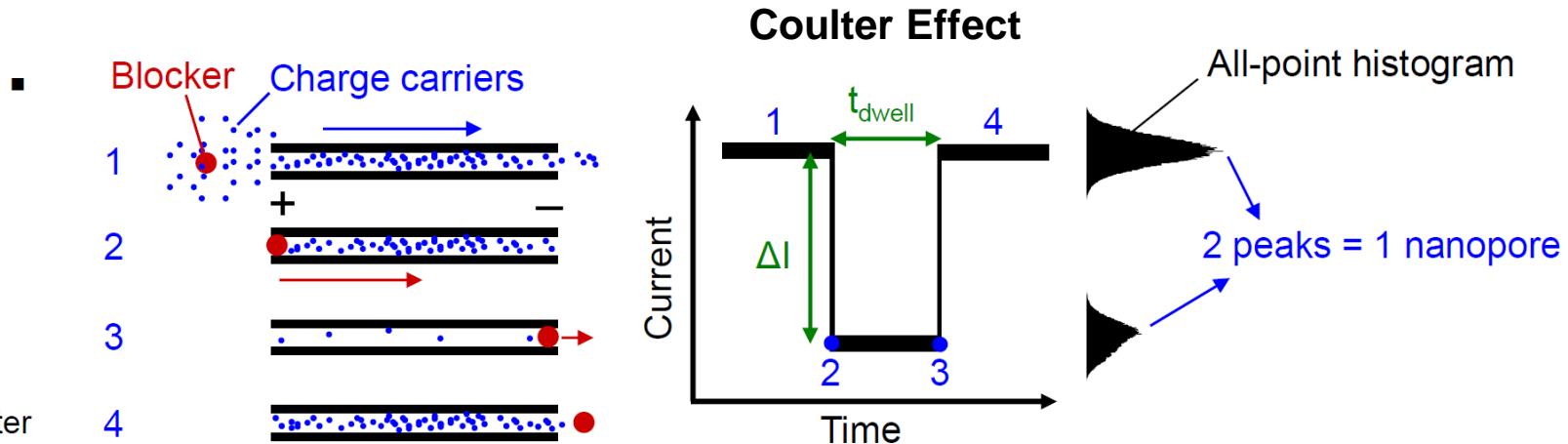
KCl 3M (with electrolyte, 1000 mV)



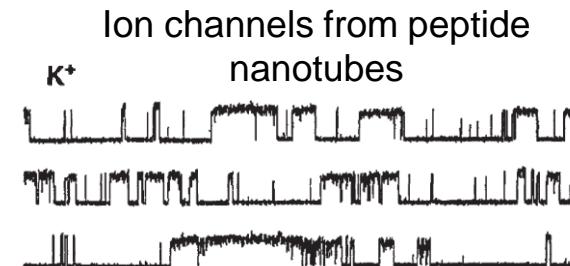
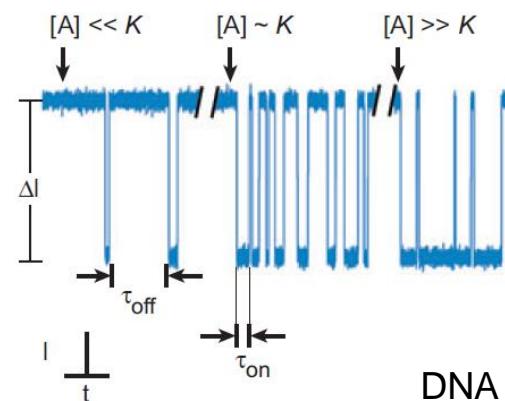
Stochastic Pore Transport: the Coulter Effect



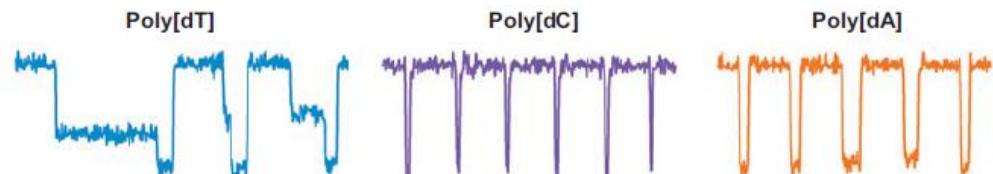
Wallace H. Coulter
(1913-1998)



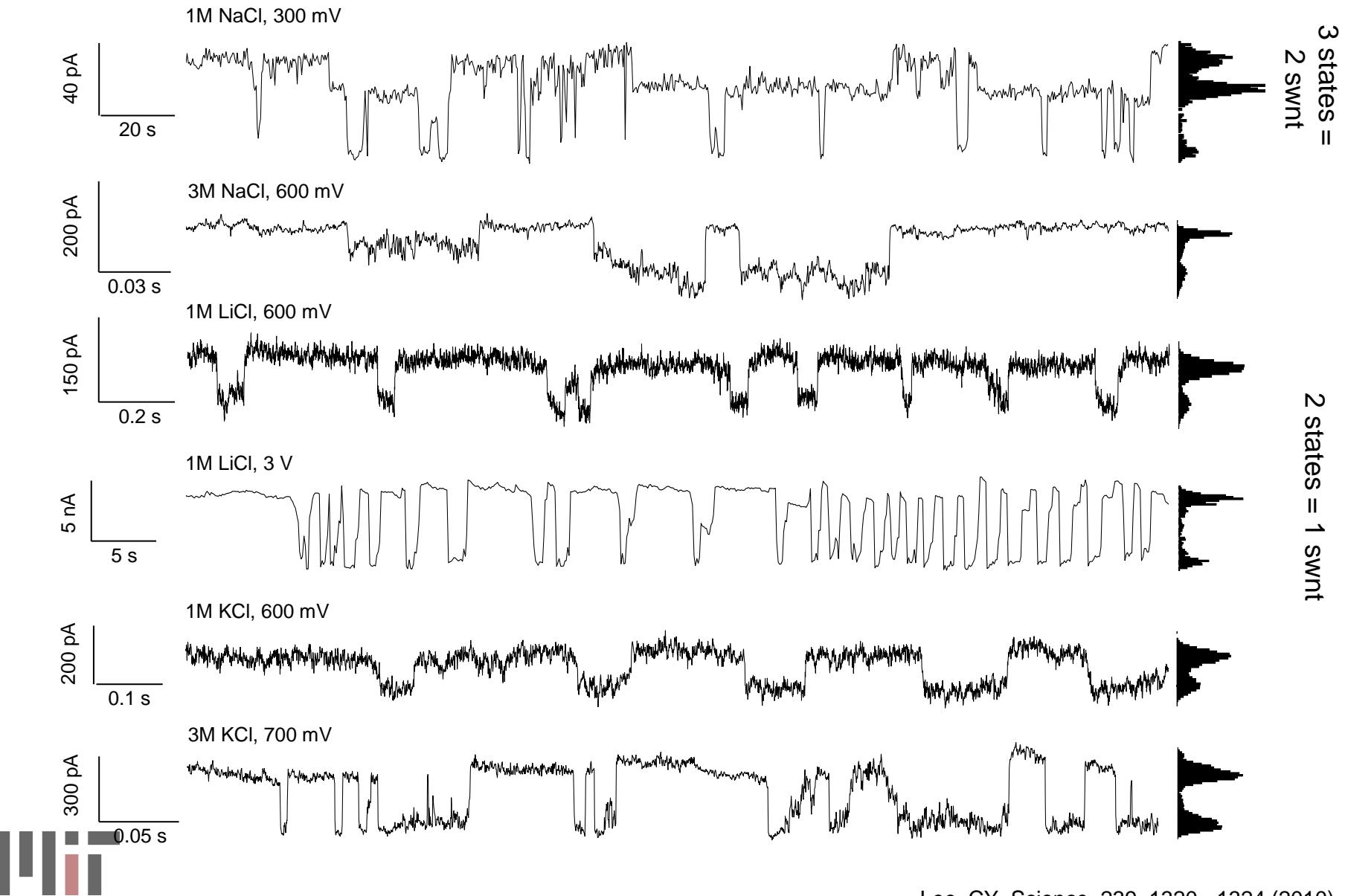
Analyte at different concentration



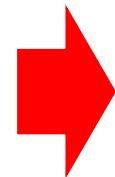
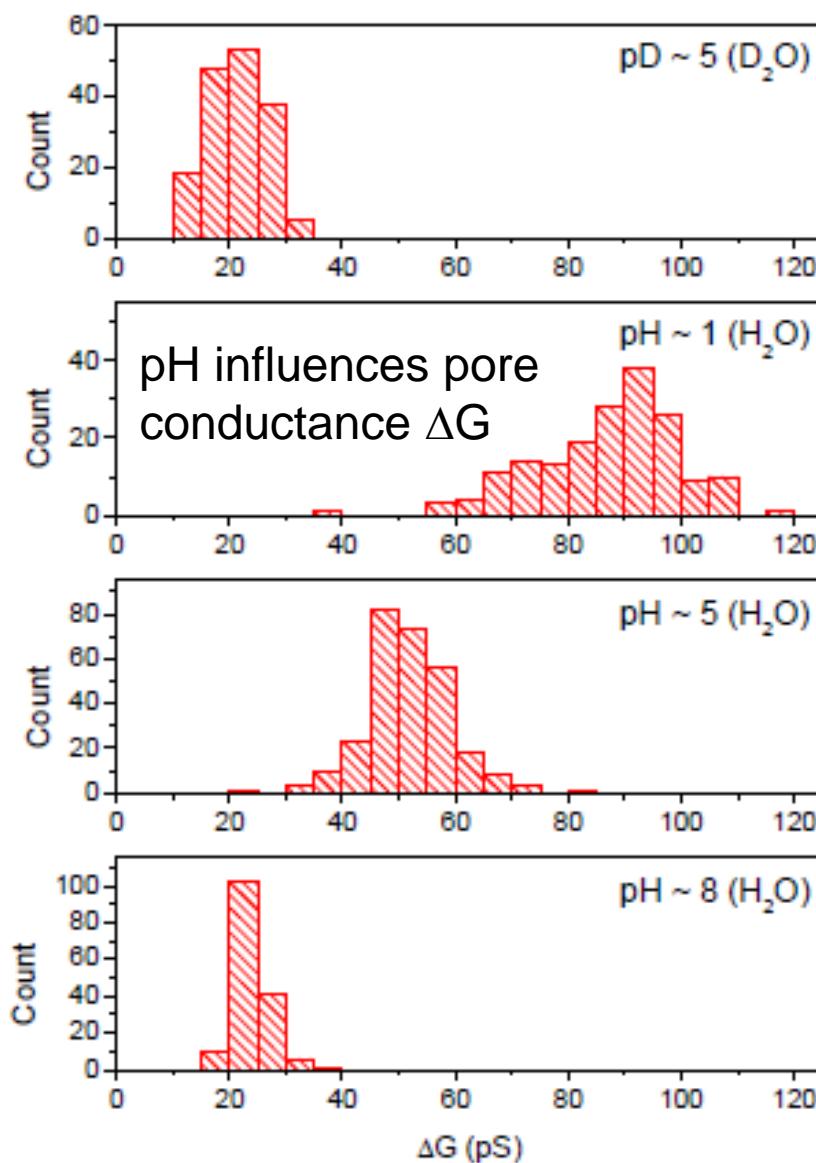
DNA sequencing by nanopore



Single Ion Transport through a Single Nanotube



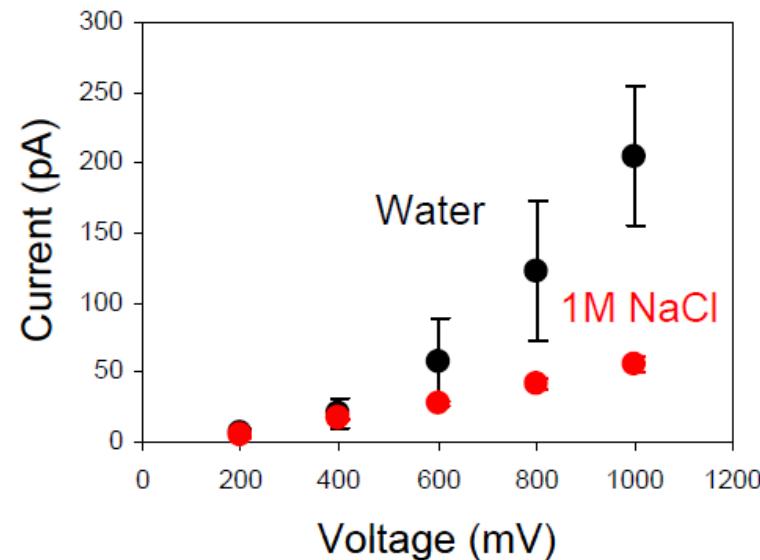
Unblocked Current is Carried by Protons



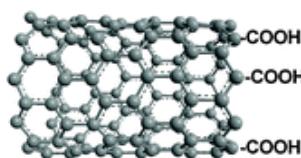
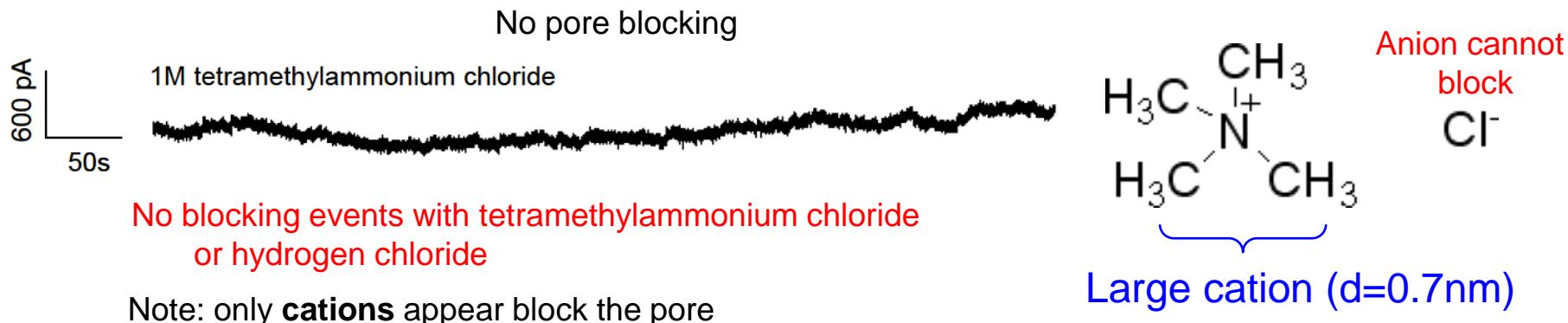
Measured $\Delta G_{H_2O} / \Delta G_{D_2O} = 2.5$

From acid/base titrations in bulk water (diffusion limited) = 1.6
(Pillings and Seakins, Reaction Kinetics, Oxford Science Press)

Cations block the nanotube

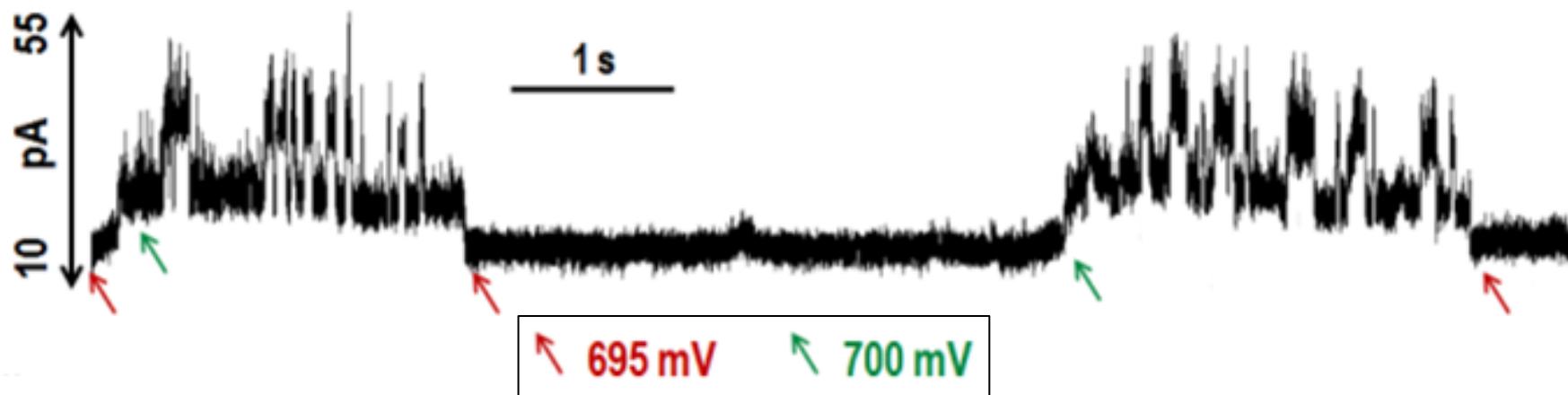


Cations are the Blocking Species



Hypothesis: plasma etch leaves residual carboxylic acid groups at pore ends (negative zeta potential)

Threshold potential observed for cation pore blocking



Theory: Molecular Transport in Nanotubes

Important theoretical/computational predictions (300+ since 2000)

Enhanced water permeation flux through carbon nanotubes

A. Alexiadis, S. Kassinos, *Chemical Reviews* **108**, 5014 (DEC, 2008).

Large proton fluxes through water filled carbon nanotubes

C. Dellago, M. M. Naor, G. Hummer, *Physical Review Letters* **90**, 105902 (Mar 14, 2003).

Ice-like water phase in carbon nanotube interior

K. Koga, G. T. Gao, H. Tanaka, X. C. Zeng, *Nature* **412**, 802 (Aug 23, 2001).

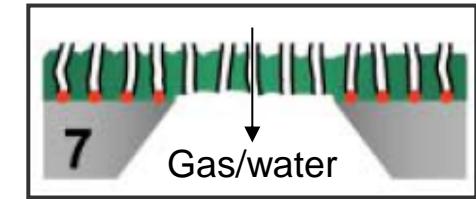
High ion rejection rates from nanotube interior

O. Beckstein, K. Tai, M. S. P. Sansom, *Journal of the American Chemical Society* **126**, 14694 (Nov 17, 2004).

Experimental: success with nanotube membranes

J. K. Holt *et al.*, *Science* **312**, 1034 (May 19, 2006).

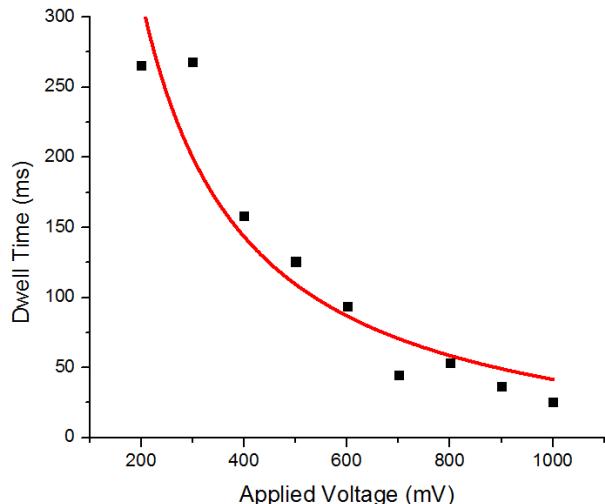
F. Fornasiero *et al.*, *Proceedings of the National Academy of Sciences of the United States of America* **105**, 17250 (Nov 11, 2008).



Holt et al. Science 312, 1034-1037 (2006)

Ion Translocation Follows Expected Nanopore Physics

- Current fluctuations for Ca^{2+} demonstrated trends expected for stochastic pore blocking



Dwell time follows simple scaling of a single ion moving through the nanotube:

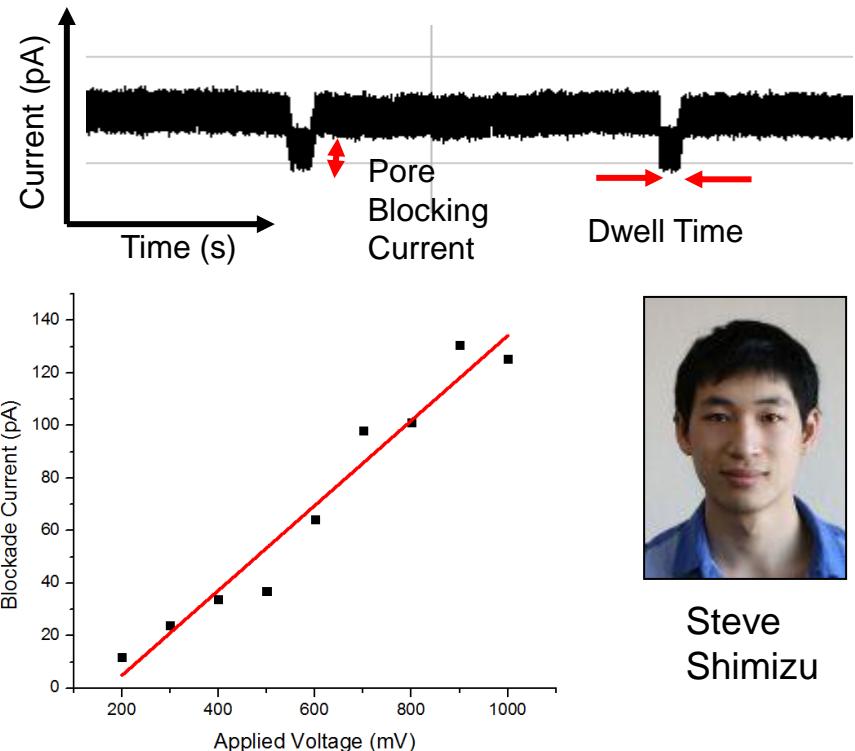
$$\tau_{dwell,ion} = \frac{L^2}{\mu_{ion} \cdot (V - V_{threshold})} \propto \frac{1}{V}$$

$\tau_{dwell,ion}$ = Dwell Time of Ion (s)

L = length of barrier (nanotube) (m)

V = applied voltage (V)

μ_{ion} = ion mobility inside nanotube ($\text{m}^2/\text{V}\cdot\text{s}$)



Pore blocking current scales linearly with applied voltage:

$$I_{blocking} = \frac{q}{\tau_{H^+}} = \frac{q \cdot \mu_{H^+} \cdot (V - V_{threshold})}{L^2} \propto V$$

$I_{blocking}$ = pore blocking current (pA)

q = charge on H^+ (C)

τ_{H^+} = average residence time of H^+ inside the nanotube (s)

μ_{H^+} = H^+ mobility inside nanotube ($\text{m}^2/\text{V}\cdot\text{s}$)

$V_{threshold}$ = threshold voltage at which pore blocking observed (V)

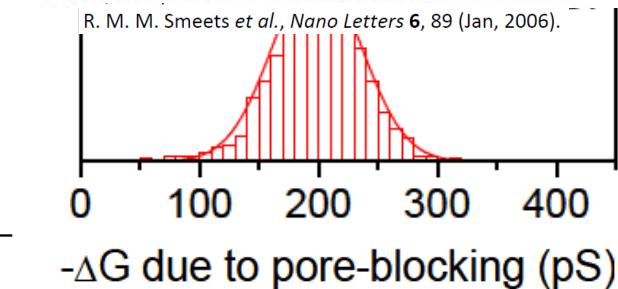
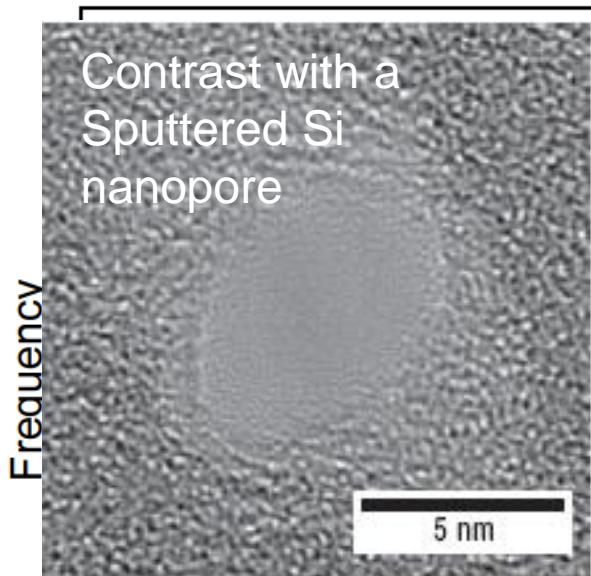
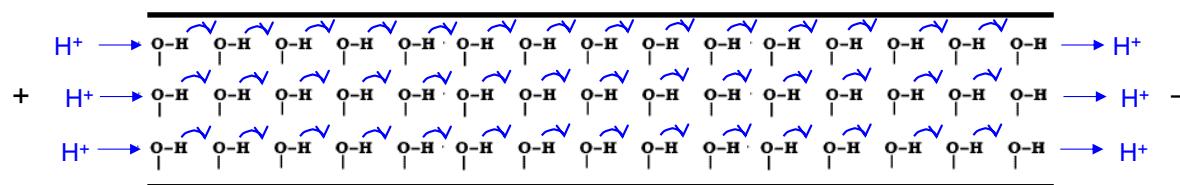
Proton Conductivity of a Nanotube

Proton conductivity is unusually large $\times 10^4$ compared with sputtered Si ion channels

Blocking current $\sim 100 \text{ pA} = 6.25 \times 10^8 \text{ protons/s}$
gramicidin (biological) proton channel $= 2.2 \times 10^9 \text{ protons/s}$ (at low pH)

Highly efficient conduction mechanism – “hop-and-turn” Grotthuss mechanism along the water chain

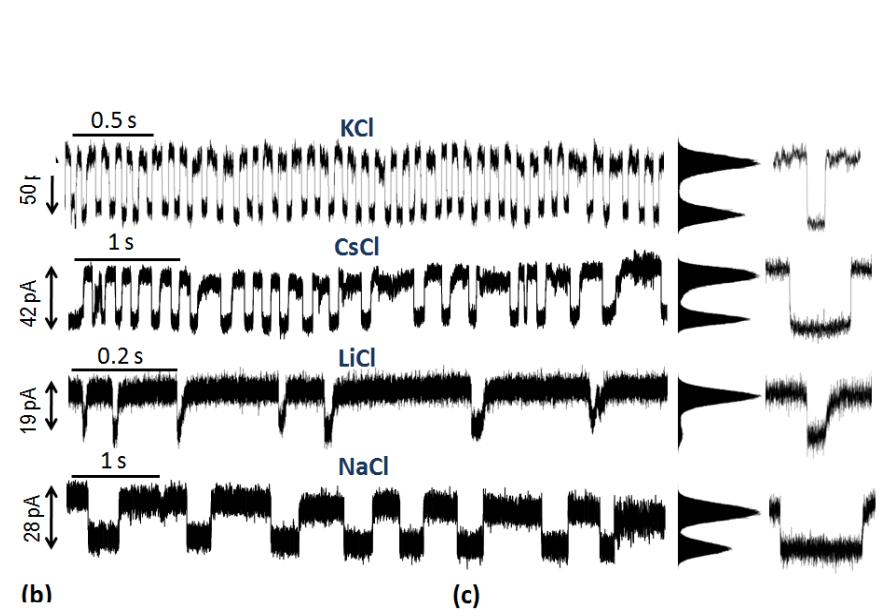
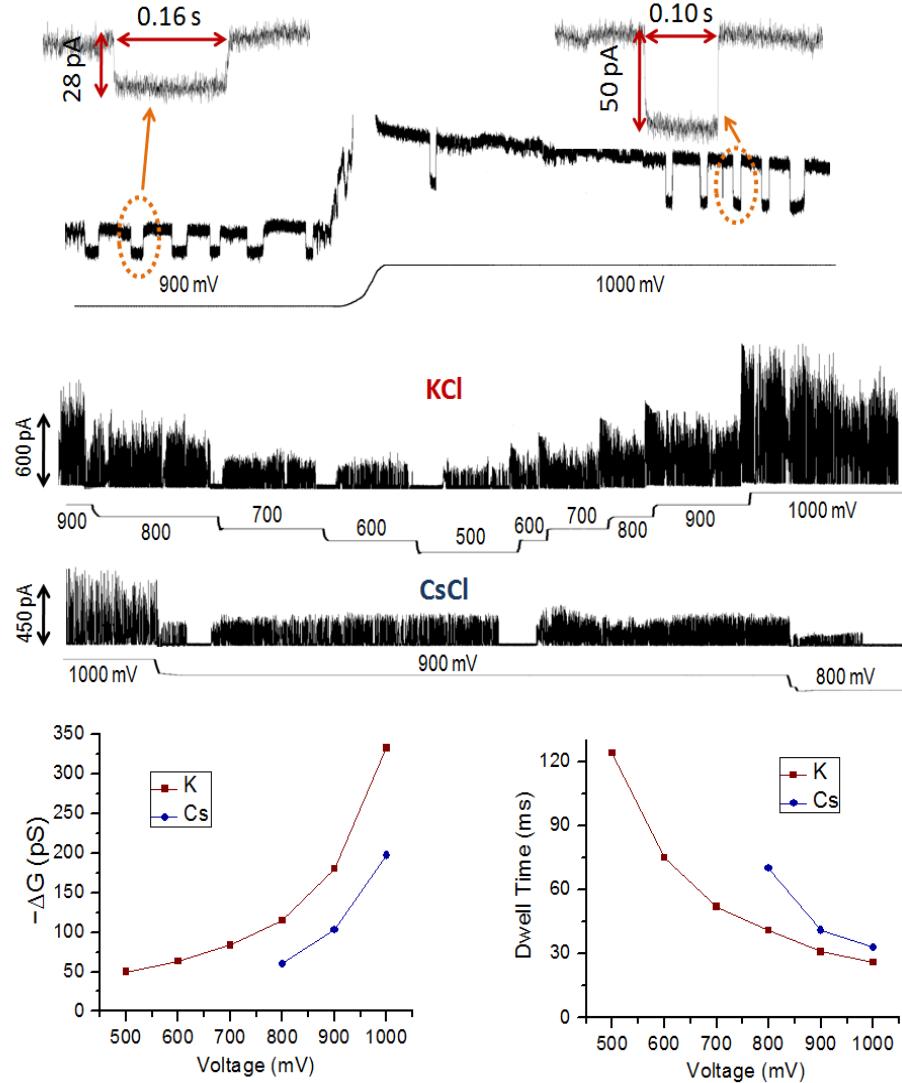
N. Agmon, *Chemical Physics Letters* **244**, 456 (Oct 13, 1995).



Divide by linear water density in tube 33/nm yields hopping velocity of 0.02 m/s and mobility of $2 \times 10^{-5} \text{ m}^2/\text{V}\cdot\text{s}$ at $E = 10^3 \text{ V/m}$ or **x 100 bulk water**.



Blockade Current and Dwell Time Dependence on Ion Type



Summary

Dwell time: $\text{Na}^+ > \text{Cs}^+ > \text{K}^+ > \text{Li}^+$

Blockade Current: $\text{K}^+ > \text{Cs}^+ > \text{Na}^+ > \text{Li}^+$

Blockade Current vs Diameter Shows Maximum

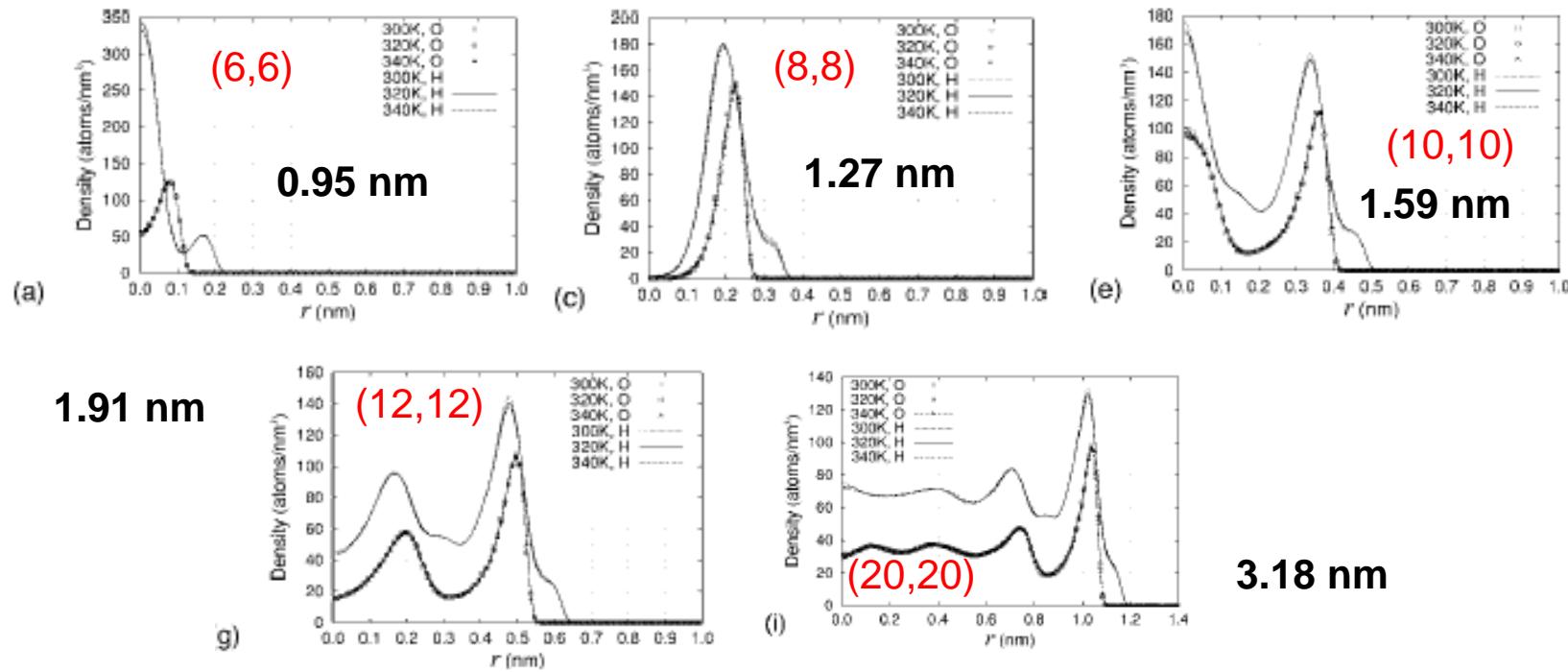
THE JOURNAL OF CHEMICAL PHYSICS 124, 174714 (2006)

Hydrogen bond dynamics and microscopic structure of confined water inside carbon nanotubes

Itsuo Hanasaki^{a)} and Akihiro Nakatani^{b)}

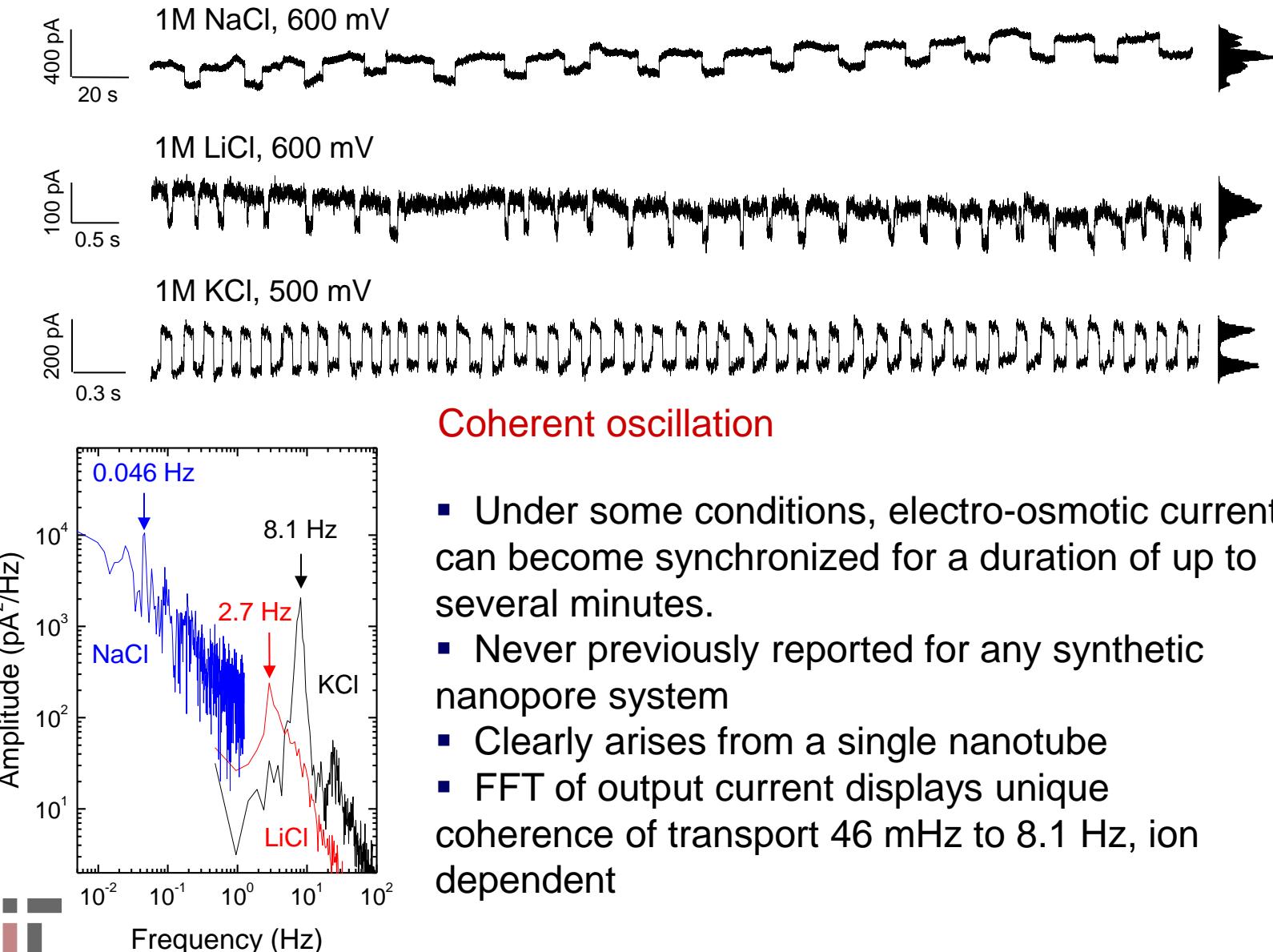
Department of Adaptive Machine Systems, Graduate School of Engineering, Osaka University,
2-1 Yamadaoka, Suita, Osaka 565-0871, Japan

(Received 28 September 2005; accepted 16 March 2006; published online 5 May 2006)



Water density increases strongly with decreasing diameter

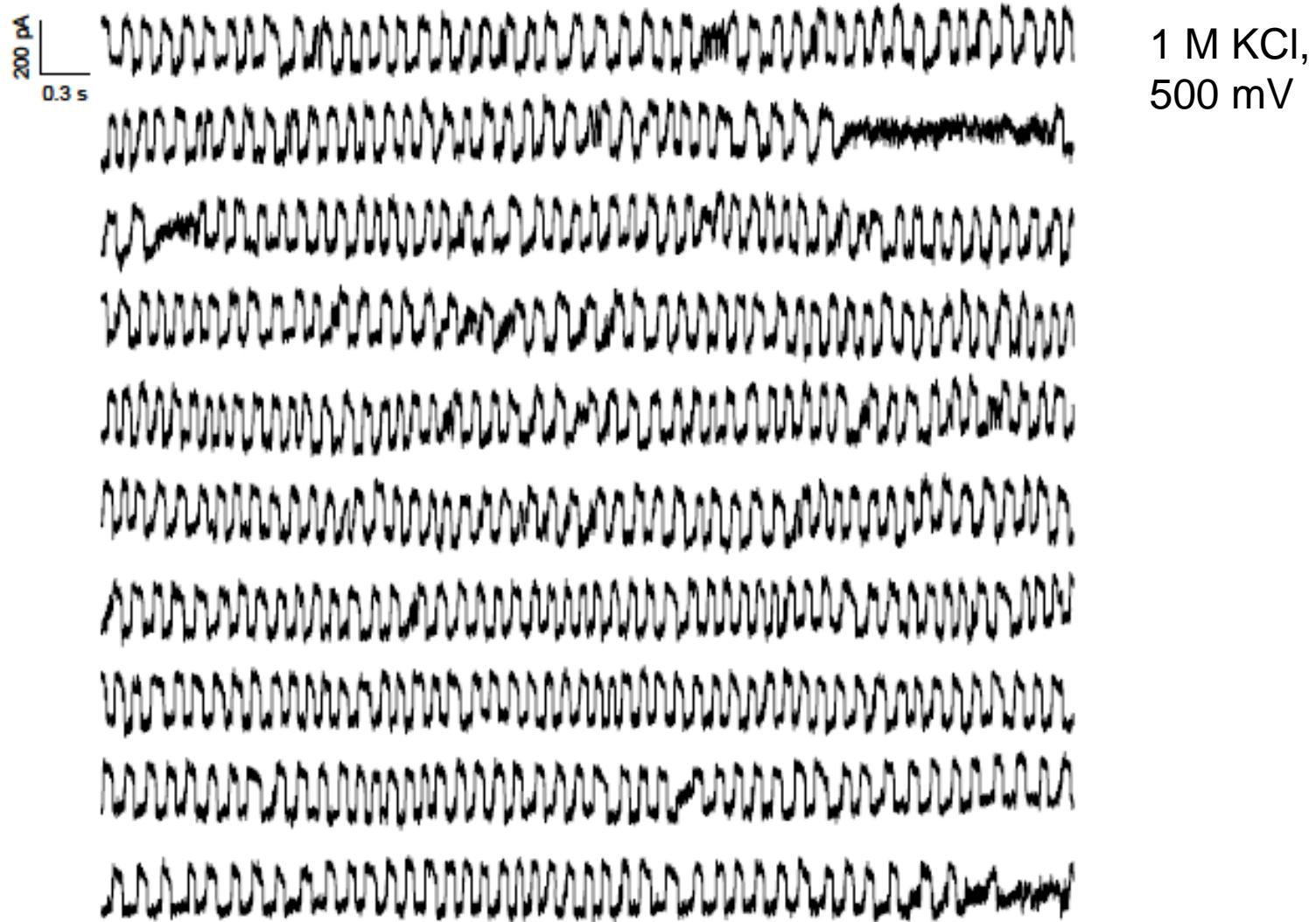
Coordinated Molecular Transport



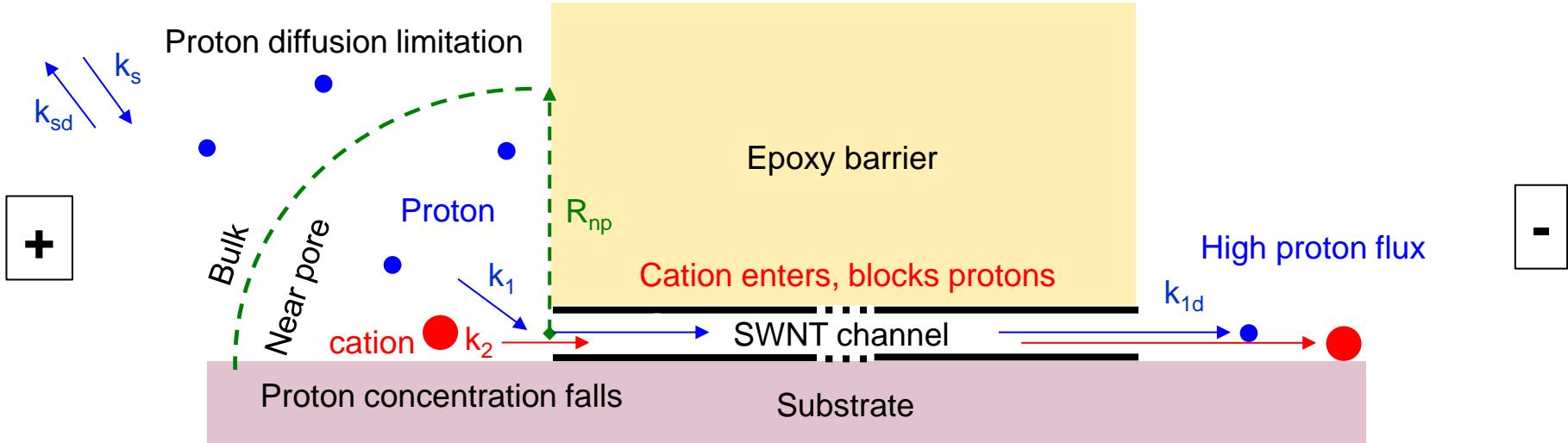
- Under some conditions, electro-osmotic current can become synchronized for a duration of up to several minutes.
- Never previously reported for any synthetic nanopore system
- Clearly arises from a single nanotube
- FFT of output current displays unique coherence of transport 46 mHz to 8.1 Hz, ion dependent



Resonant, Synchronized Transport

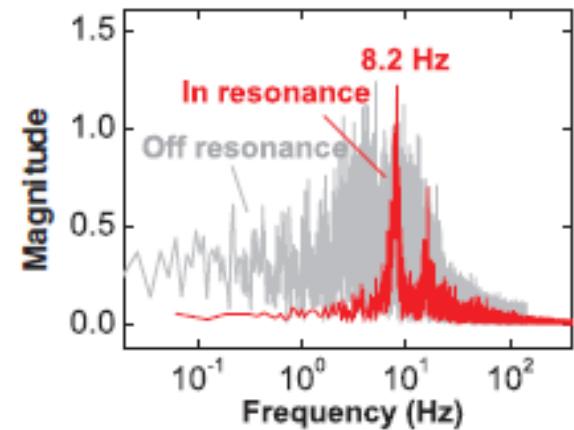


Stochastic Simulation Explains the Oscillation



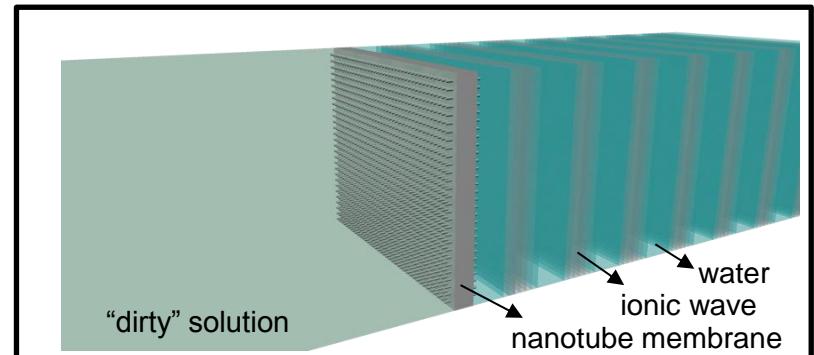
Mechanism can be simulated via 6 stochastic differential equations

- 1) $H_{np} + T \xrightarrow{k_1} H_{tube}$ (proton in)
 - 2) $I_{np} + T \xrightarrow{k_2} I_{tube}$ (ion in)
 - 3) $H_{tube} \xrightarrow{k_{1d}} \text{(proton out)}$
 - 4) $I_{tube} \xrightarrow{k_{2d}} \text{(ion out)}$
 - 5) $H_{bulk} \xrightarrow{k_s} H_{np}$ (proton exchange)
 - 6) $H_{np} \xrightarrow{k_{sd}} H_{bulk}$
- $T = \text{open nanopore state}$



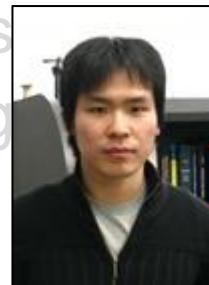
Summary and Future Work – Ion Transport

- Carbon nanotube: new **experimental conduit** to manipulate single molecules in a nano-confined channel; test computational predictions
- **Coherence resonance** – dramatically increases the throughput of a nanopore sensor (x 100)
- Confined reactions and reactors, molecular synthesis and catalysis **one molecule at a time**
- Trace ion detection and ion separation in aqueous phases



New Concepts in Mass and Energy Transport within Carbon Nanotubes

- Coherence resonance in molecular transport through a single walled carbon nanotube nanopore
- Energy storage and generation using **thermopower waves**
- Near infrared fluorescent sensors for oxygen molecule sensitivity for studying biological signaling molecules such as reactive oxygen and nitric oxide signaling in the Epidermal Growth Factor Receptor (EGFR)



Won Joon
Choi



Joel
Abrahamson

Chemical Energy Densities are Much Larger than Electrochemical for Battery Applications

Electrochemical

Energy Density (MJ/L)

<u>Li-ion</u>	1.44
<u>NiCd</u>	0.72
<u>NiMH</u>	1.08



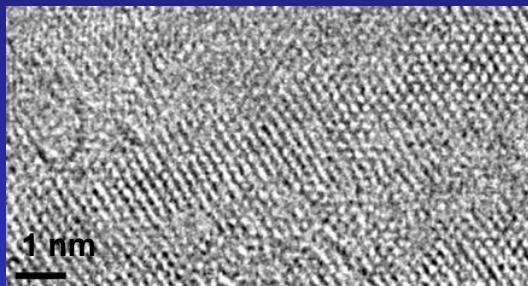
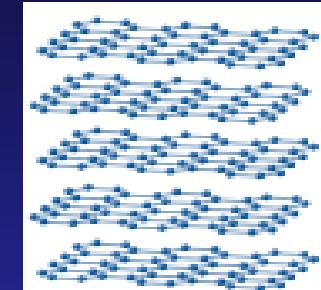
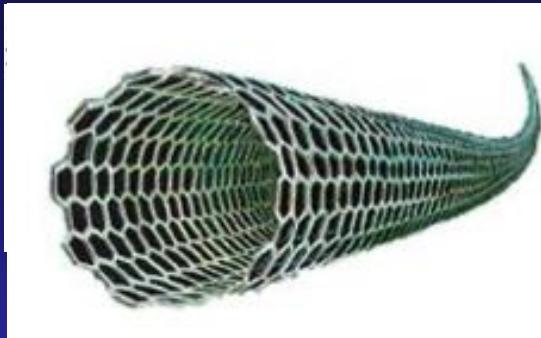
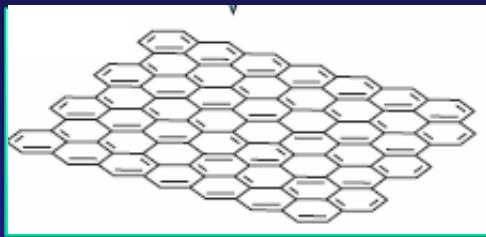
Chemical

<u>Ethanol</u>	21.2
<u>Glucose</u>	24.0
<u>Methane</u>	22.2

Efficient, direct conversion of **chemical** to **electrical** energy is an unresolved problem.

Nanomaterials can demonstrate large anisotropic thermal conductivities

Quantum confinement suppresses phonon scattering mechanisms



Graphene

2D confined material
Thermal conductivities
3080 to 5150 W/m/K in plane

J. C. Meyer, et al. *Nature*. 2007, 446
A. A. Balandin et al. *Nano Lett.* 2008, 8

Single Walled Carbon Nanotube

1D confined material
Thermal conductivities
2,500 to 10,000 W/m/K axial

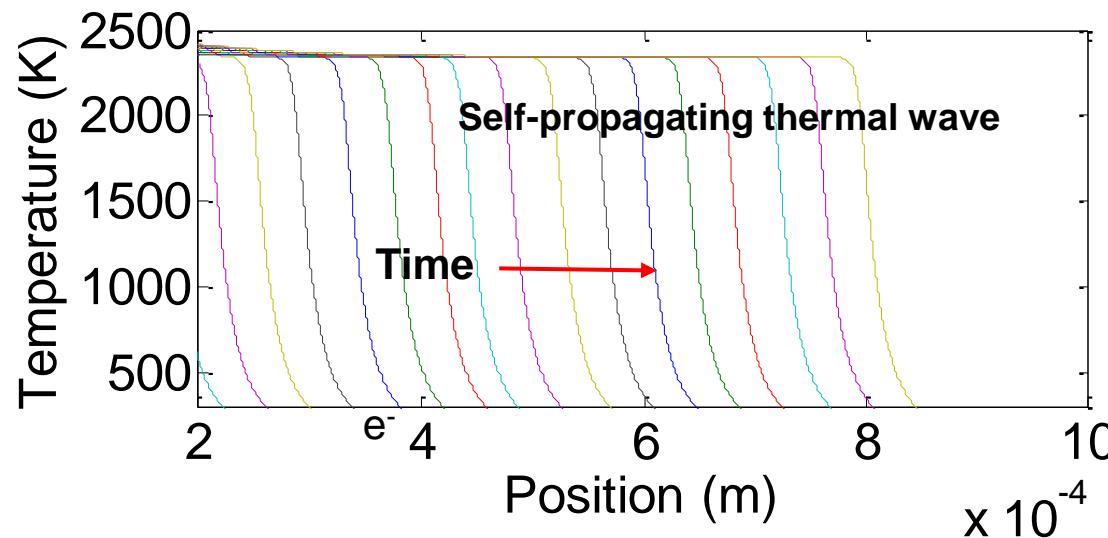
C. H. Yu et al. *Nano Lett.*, 2005, 5
E. Pop et al. *Nano Lett.*, 2006, 6

Graphite

3D bulk material
Thermal conductivities
5.7 (\perp) to 1950 W/m/K (\parallel)

Handbook of Chem. and Phys., 2010
B. T. Kelly, *Physics of Graphite*, 1981.

Thermopower wave: reaction wave pushes electrons



Heat transfer from nanowire to fuel Fuel

$$\frac{\rho C_p}{M_w} \frac{\partial T}{\partial t} = \chi \frac{\partial^2 T}{\partial x^2} + Q \left(\frac{\partial y}{\partial t} \right) - G(T - T_2)$$

enthalpy of reaction Molar density of fuel (1st order decomposition)

molar weight thermal conductivity

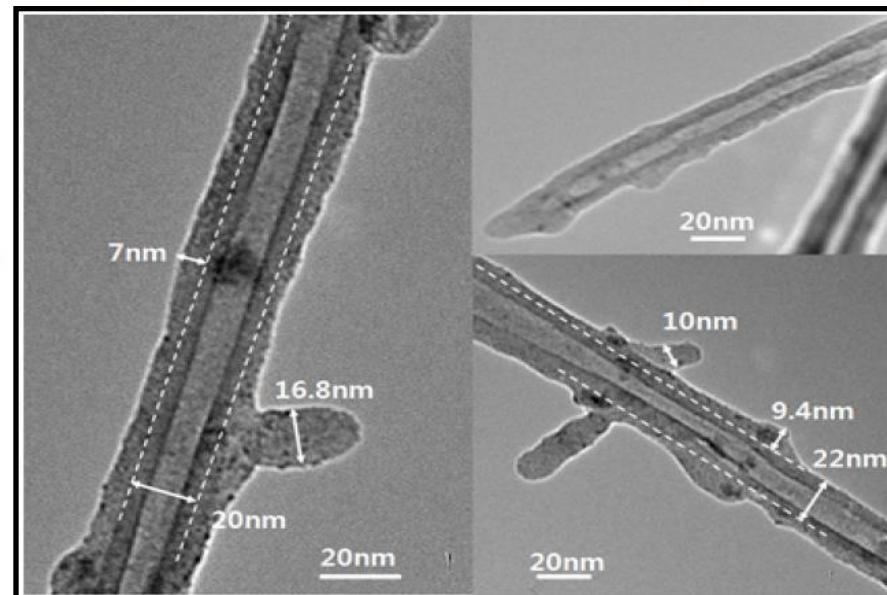
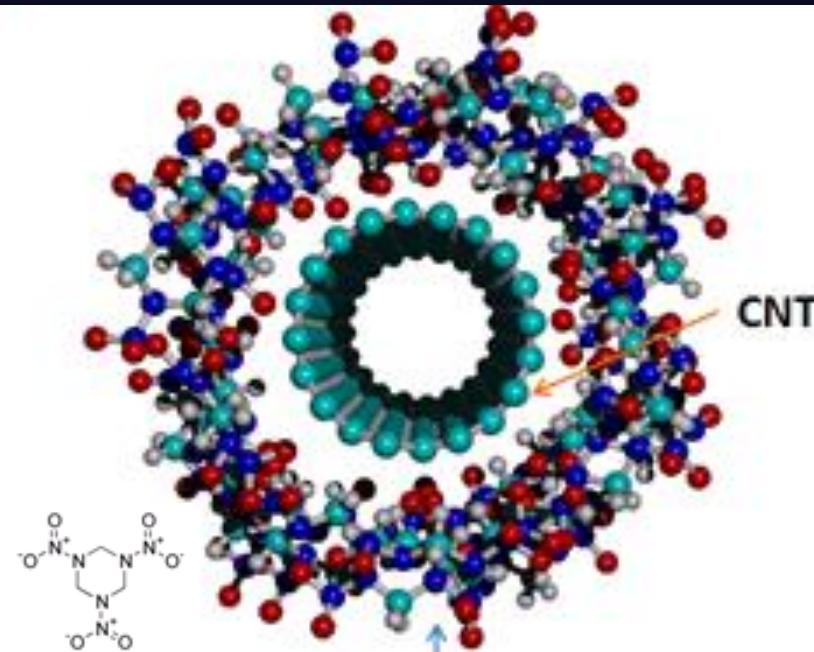
interfacial conductivity

Energy balance for fuel

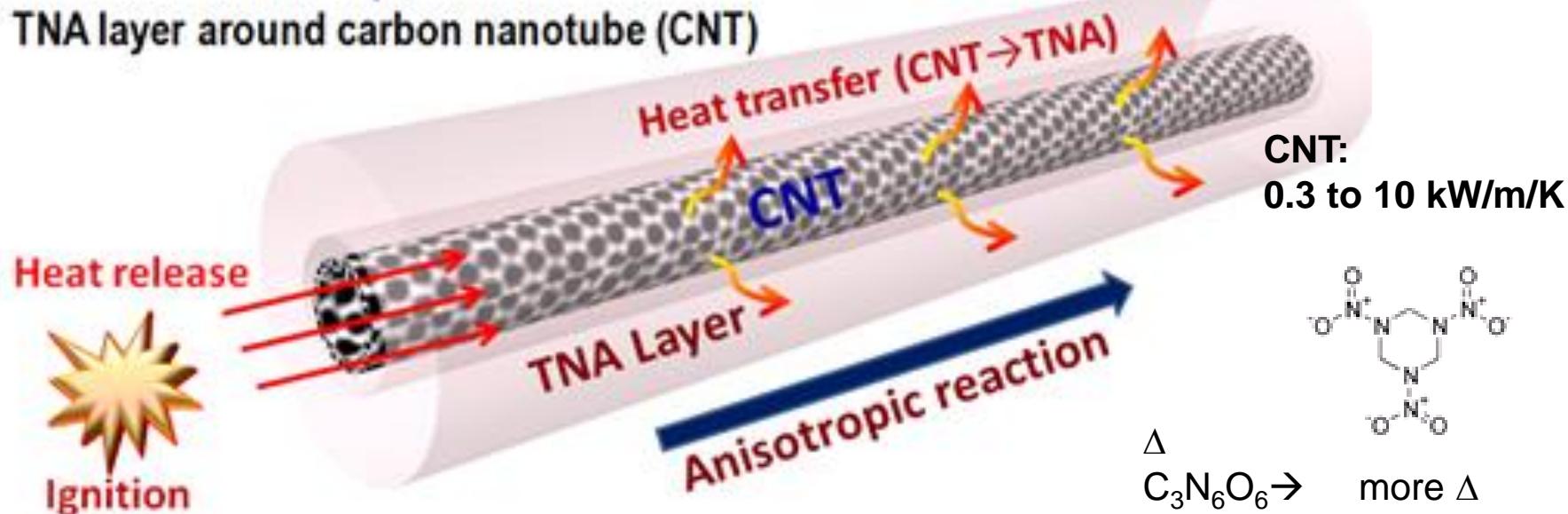
$$\frac{\rho_2 C_{p,2}}{M_{w2}} \frac{\partial T_2}{\partial t} = \chi_2 \frac{\partial^2 T_2}{\partial x^2} + G(T - T_2)$$

Energy balance for thermal conduit

Chemically driven thermal waves along CNT

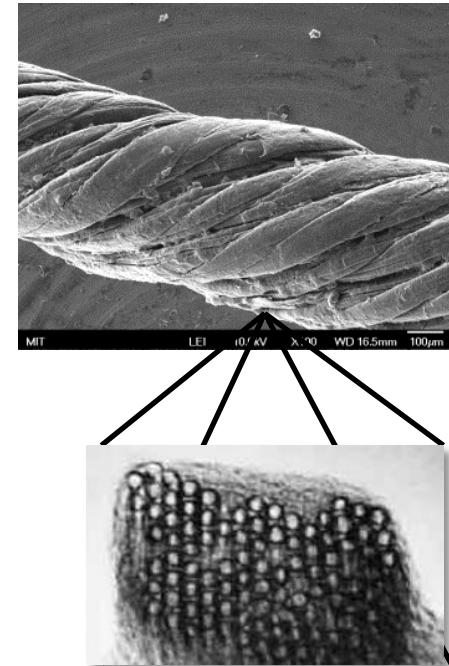
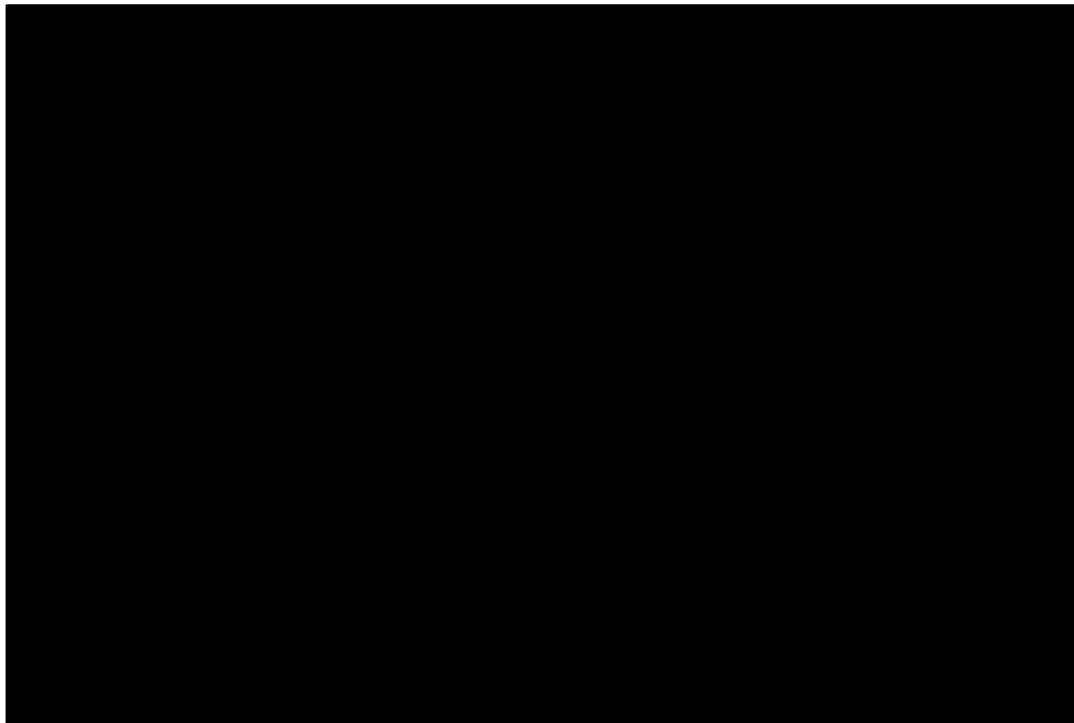


TNA layer around carbon nanotube (CNT)



Thermopower Waves in Carbon Yarns

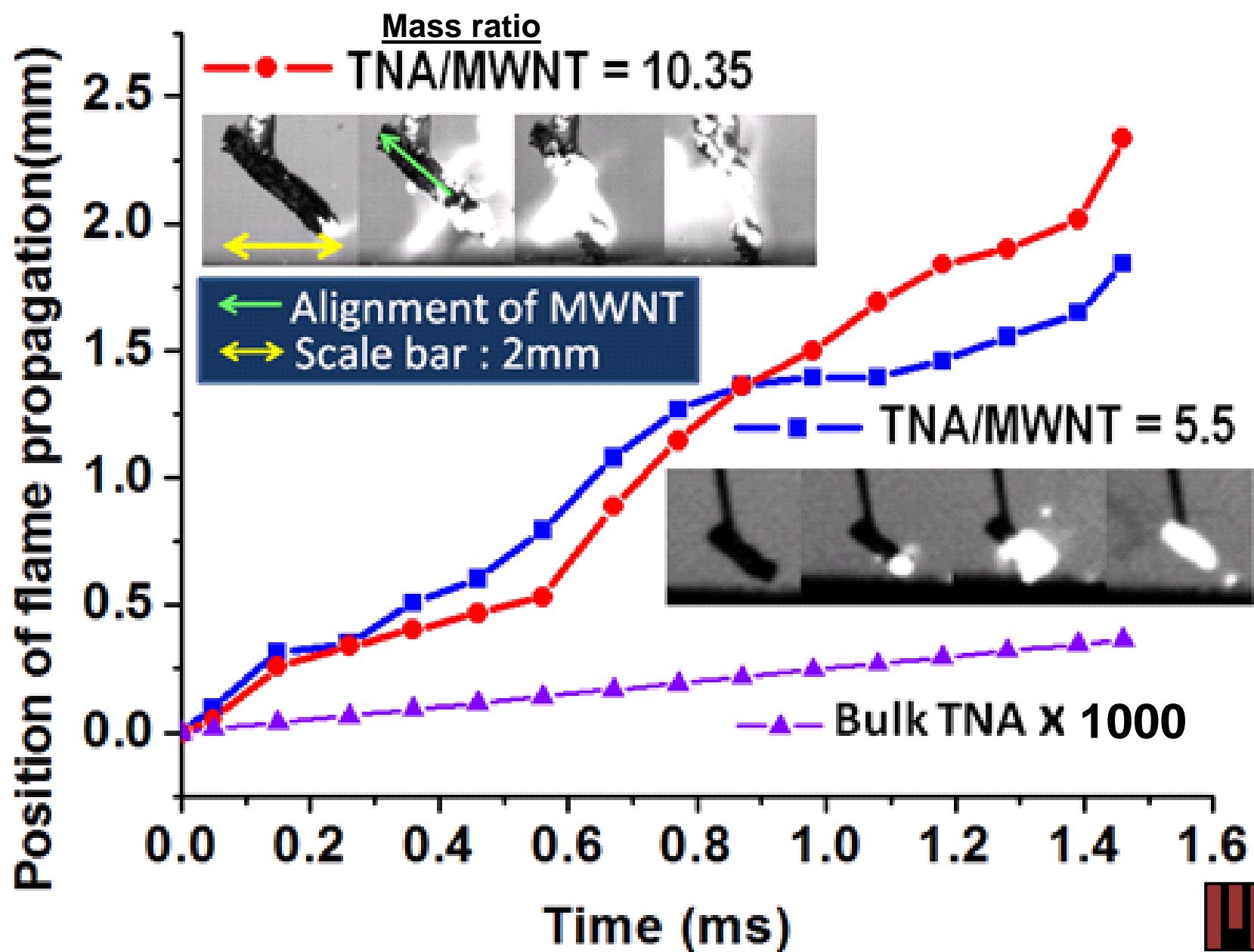
Carbon nanotubes can be spun into **yarns**,
and coated with organic fuel (trinitroamine)



Massachusetts Institute of Technology

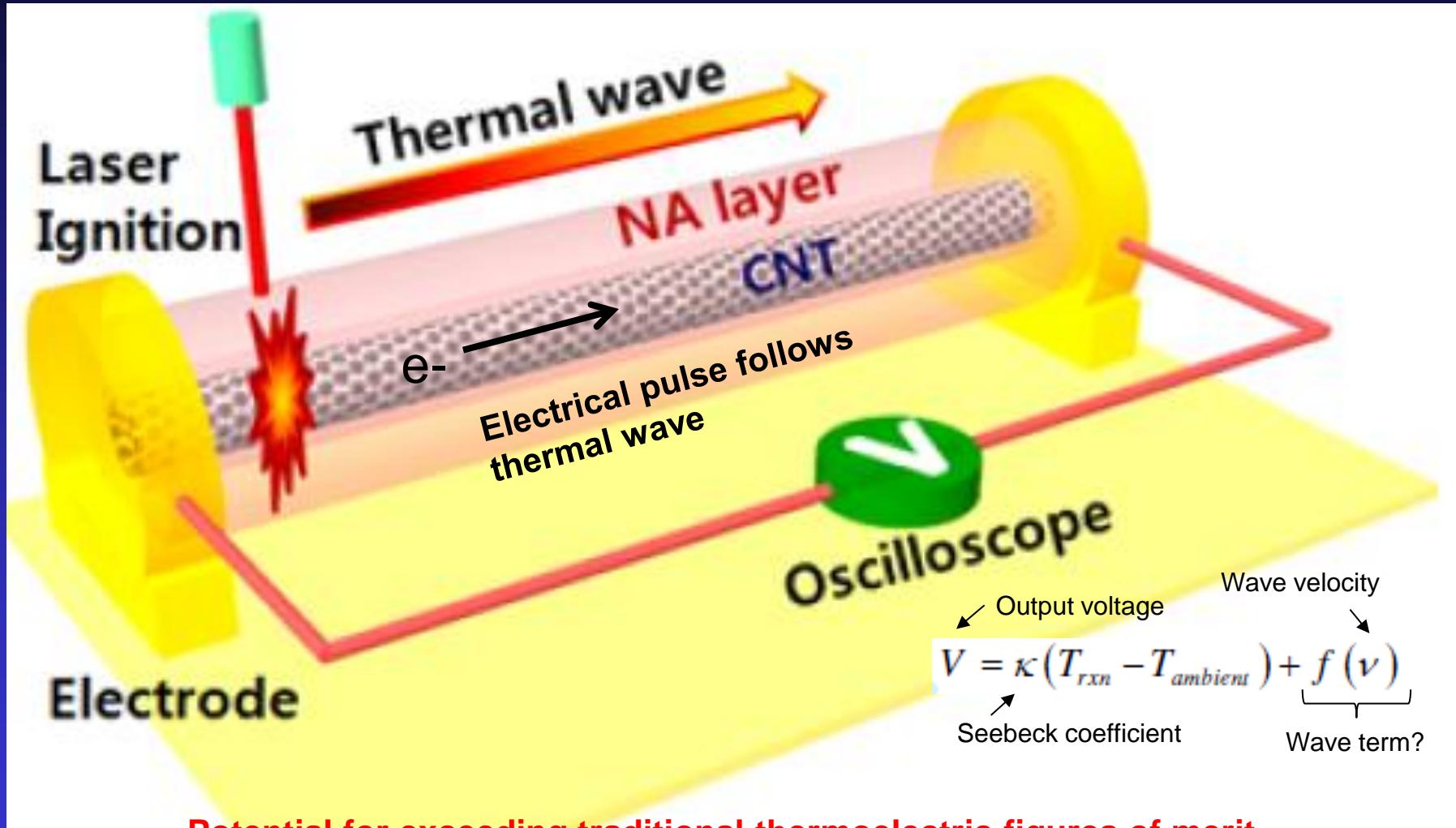
Slide 34

1D thermal wave is accelerated x10000



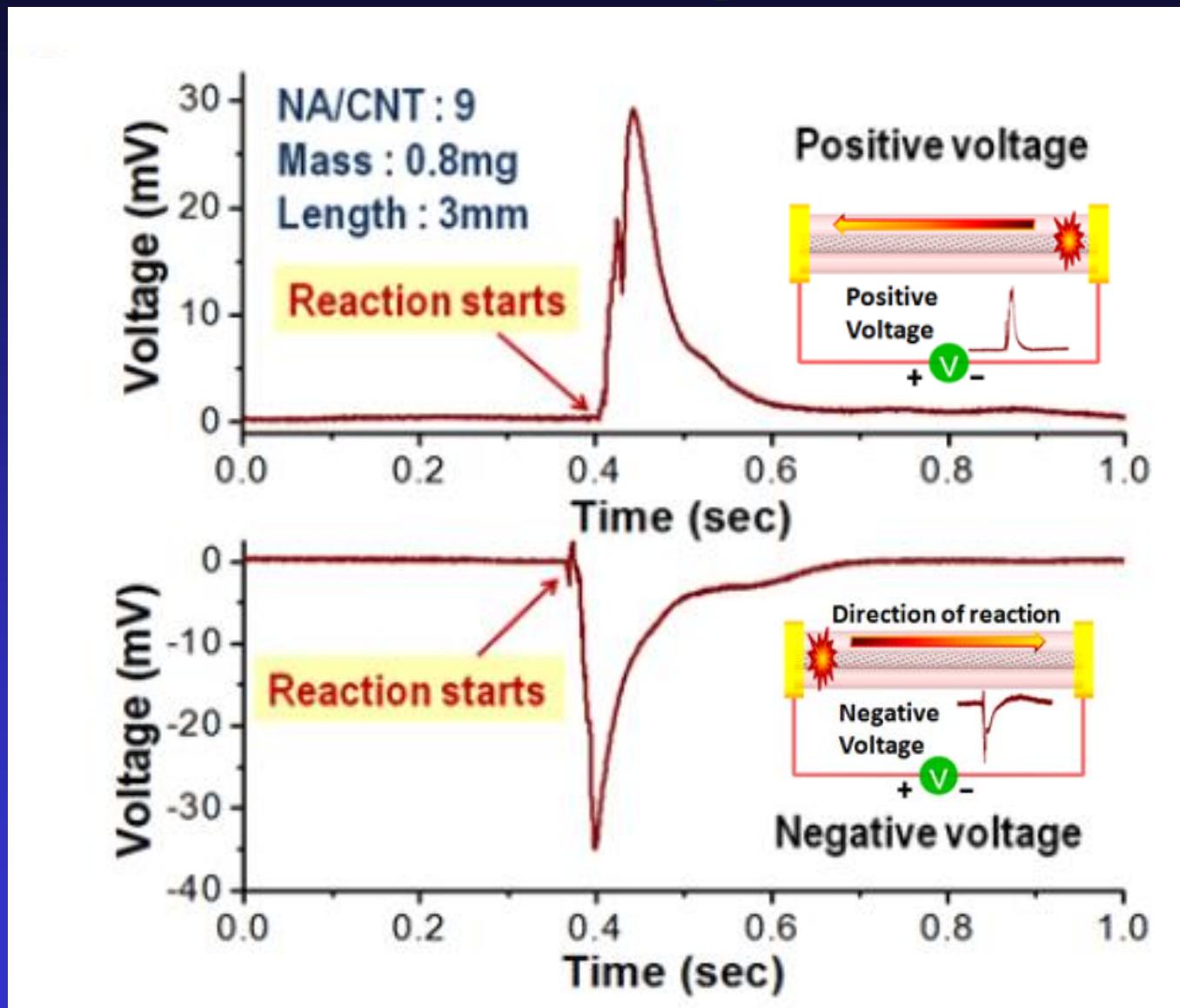
Reaction induces an electrical pulse: ***thermopower wave – a first demonstration***

Electrons entrained in the thermal wave drive a current along nanotube

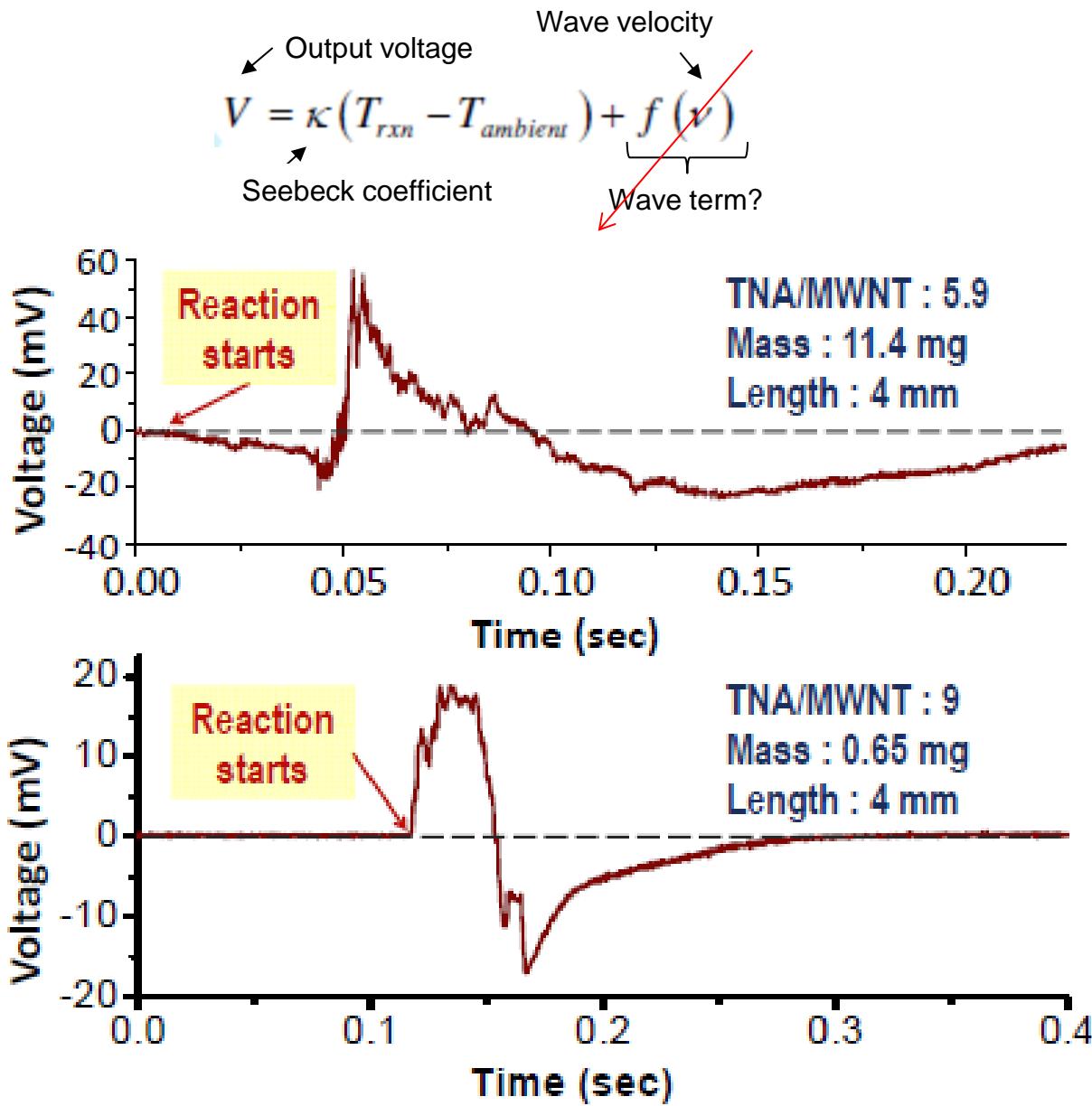


Potential for exceeding traditional thermoelectric figures of merit
Thermal gradient is preserved at the reaction front

Thermopower electrical pulse follows thermal wave propagation



Slow Wave Velocities Yield Conventional Thermopower



Conventional thermopower (a slow wave) yields a sinusoidal voltage profile as the reaction zone traverses the nanotube



Thermopower Wave Research at MIT

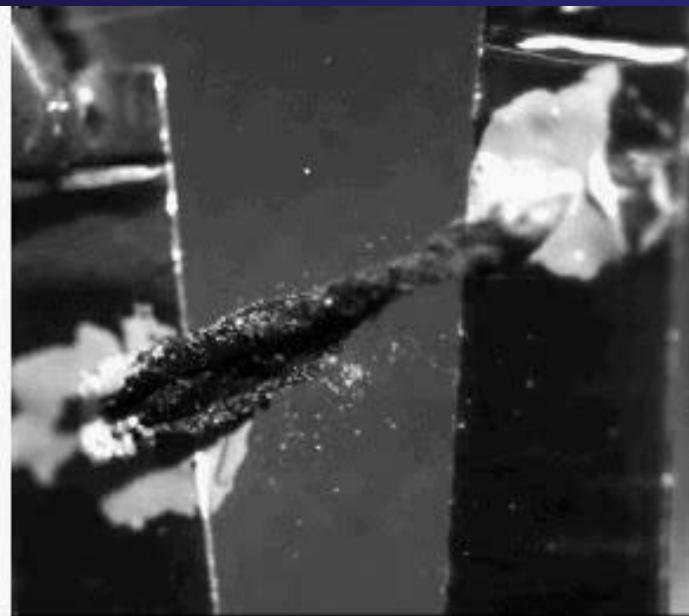
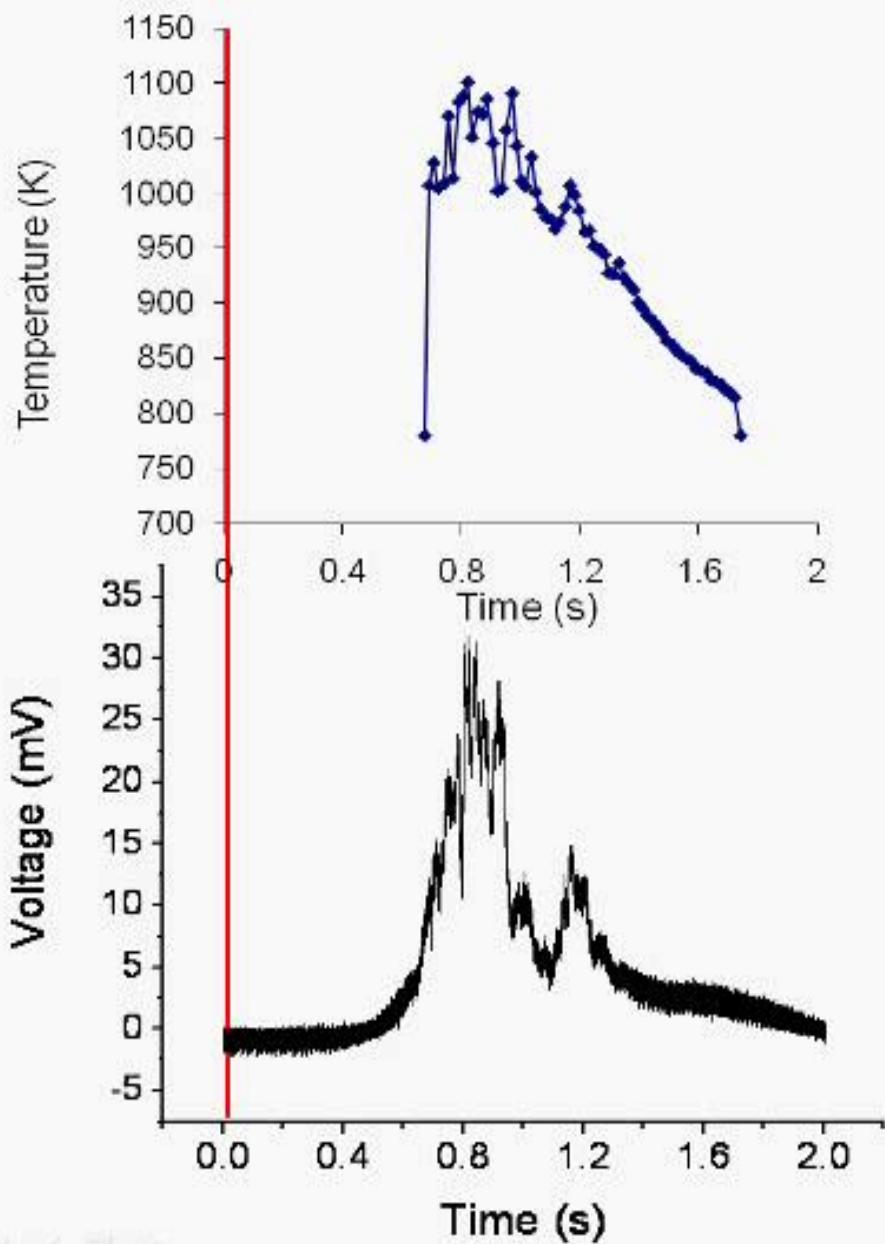


Massachusetts Institute of Technology

Courtesy: The Discovery Channel

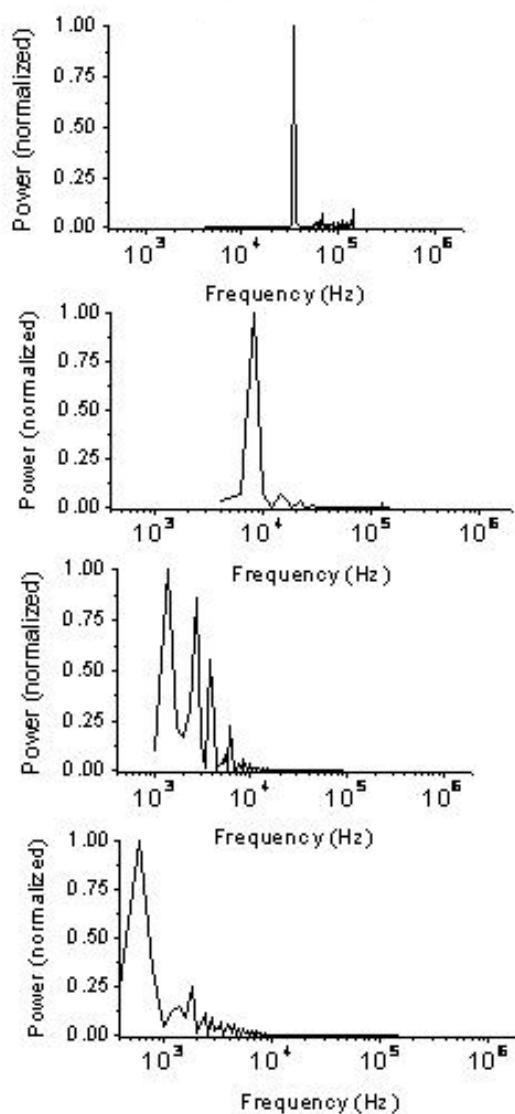
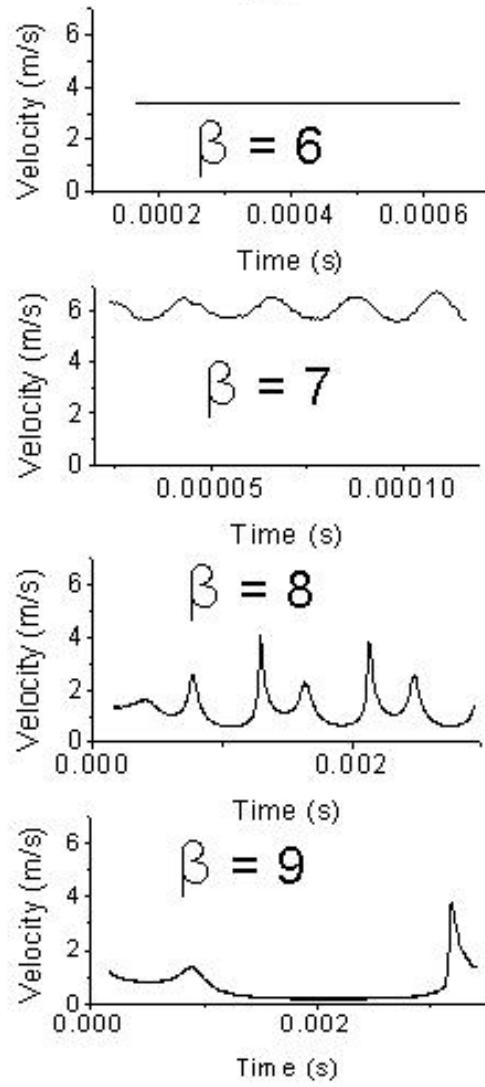
Slide 39

Temperature, Voltage and Visual Sync



The carbon survives the high temperature and can be used again!

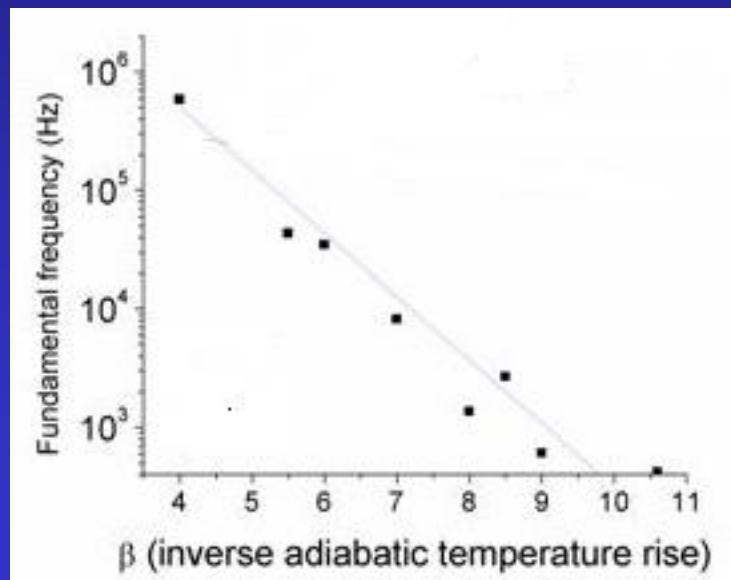
Theoretical Calculations Predict Oscillating Thermal Wavefronts: AC Batteries



$$\beta = \frac{C_p E_a}{-QR}$$

Annotations for the equation:

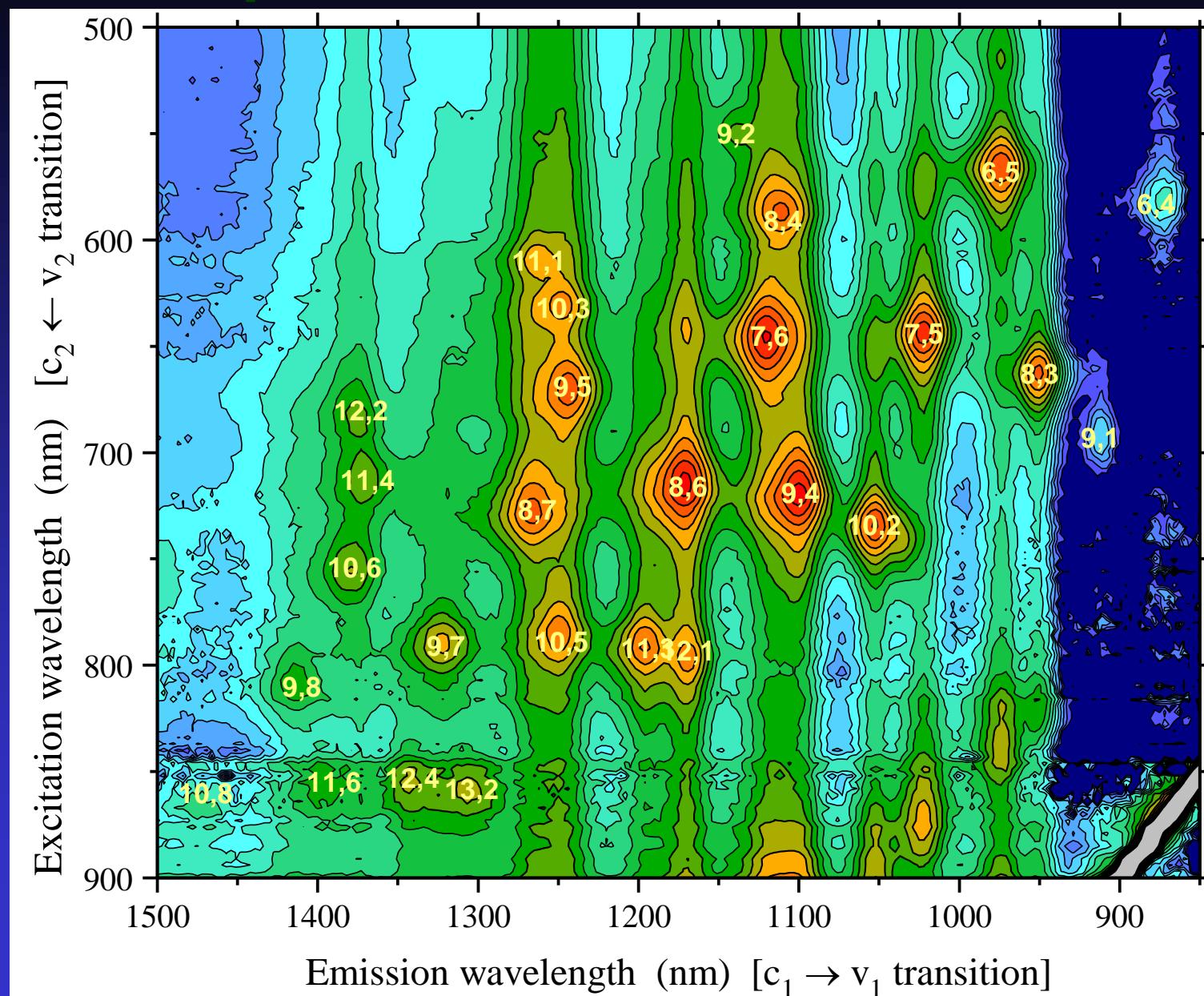
- Heat capacity: C_p
- Activation energy: E_a
- Inverse adiabatic reaction temperature: $-QR$
- Heat of reaction: Q



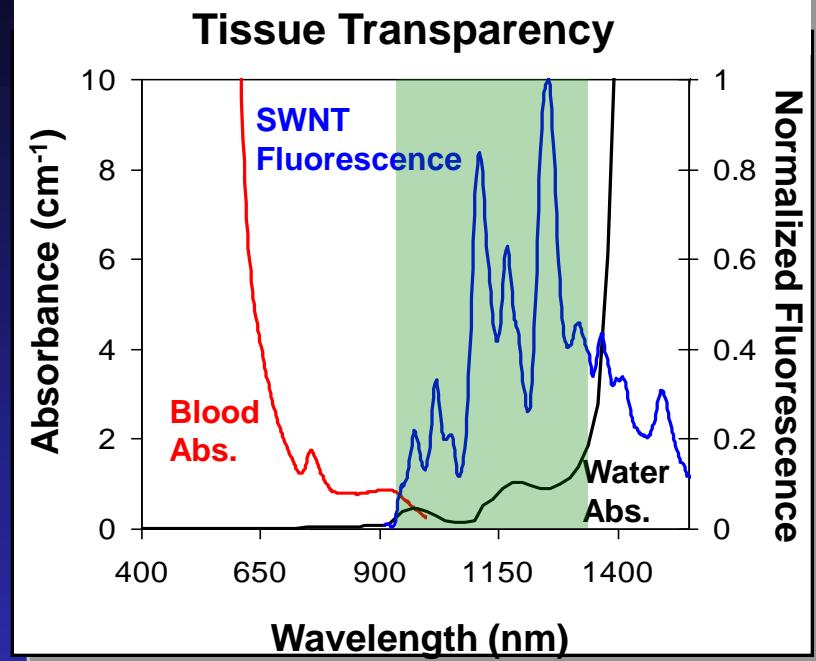
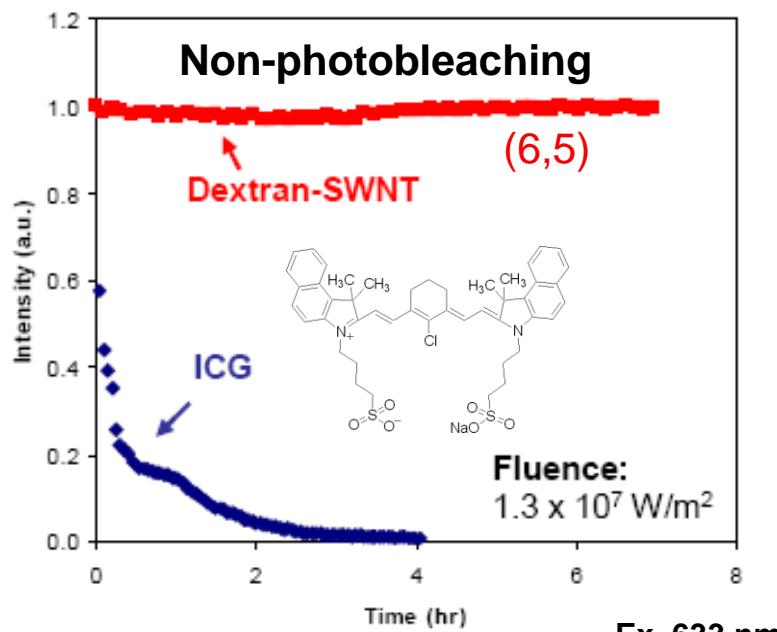
New Concepts in Mass and Energy Transport within Carbon Nanotubes

- Coherence resonance in molecular transport through a single walled carbon nanotube nanopore
- Energy storage and generation using thermopower waves
- Near infrared **fluorescent sensors** with single molecule sensitivity for studying biological signaling fluxes: reactive oxygen and nitric oxide signaling in Epidermal Growth Factor Receptor (EGFR)

Optical Properties: Near Infrared Band Gap



Semiconducting Carbon Nanotubes Fluoresce



SWNT are non-photo-bleaching, no blinking at high fluence

Near infrared emission is uniquely situated in the *tissue transparency window*; no auto-fluorescence from biomaterials

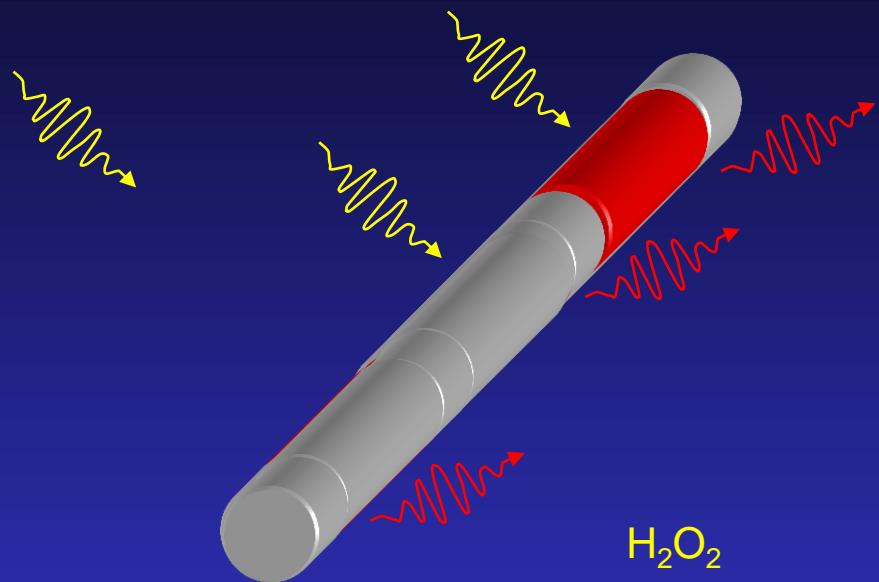
No surface states, environmentally sensitive (two types):

Solvatochromism (emission wavelength changes)

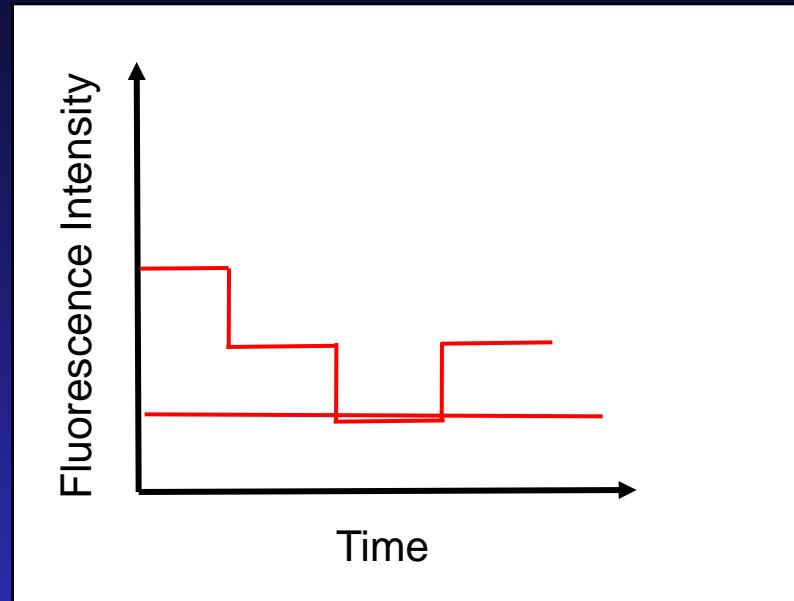
Charge transfer, quenching (intensity variation)

Single Molecule Detection using SWNT Fluorescence

1D confined exciton can detect stochastic adsorption of single molecules



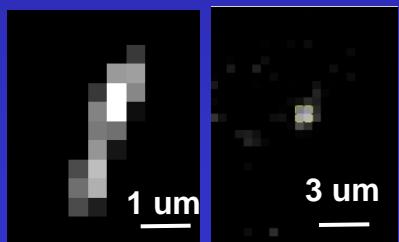
H₂O₂



Exciton excursion distance = 90 nm Cognet et al *Science* 316, 1465-1468 (2007)

In diffraction limited spot, as many as 900 nm/ 90 nm ~ 10 emission states (3 shown above)

Quenching molecules (NO, H₂O₂, H⁺, Fe(CN)₆³⁻) extinguish emission of state stochastically
Jin et al. *Nanoletters* 8 (12) 4299-4304 (2008)



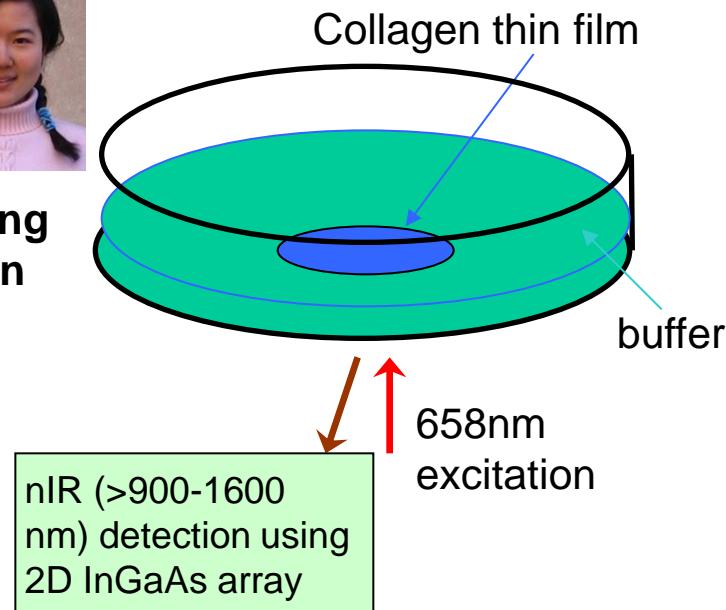
Long SWNT

Typical SWNT

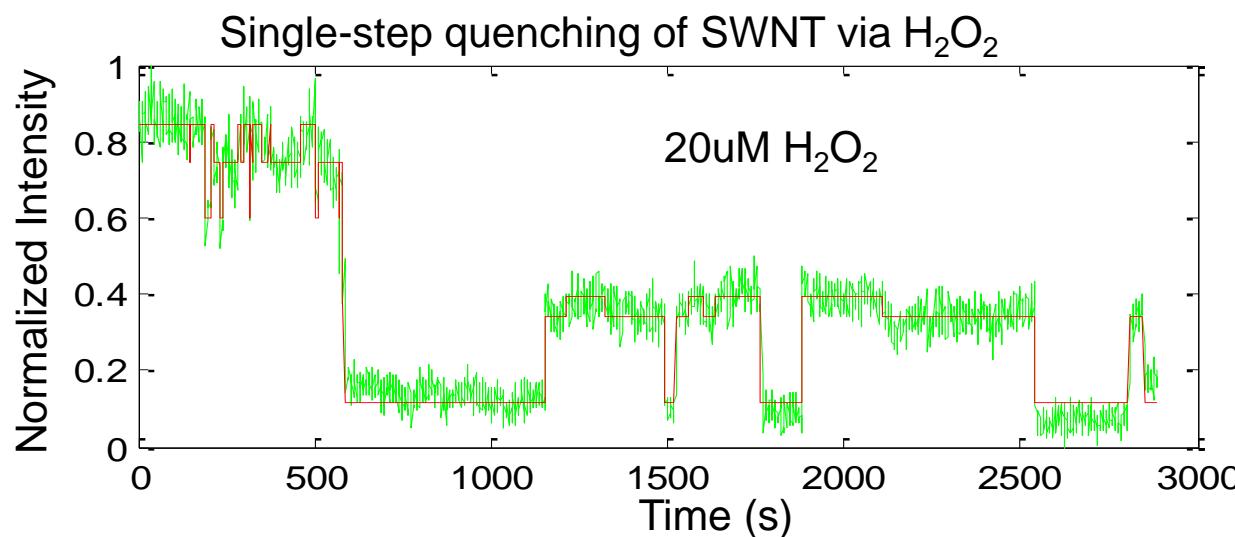
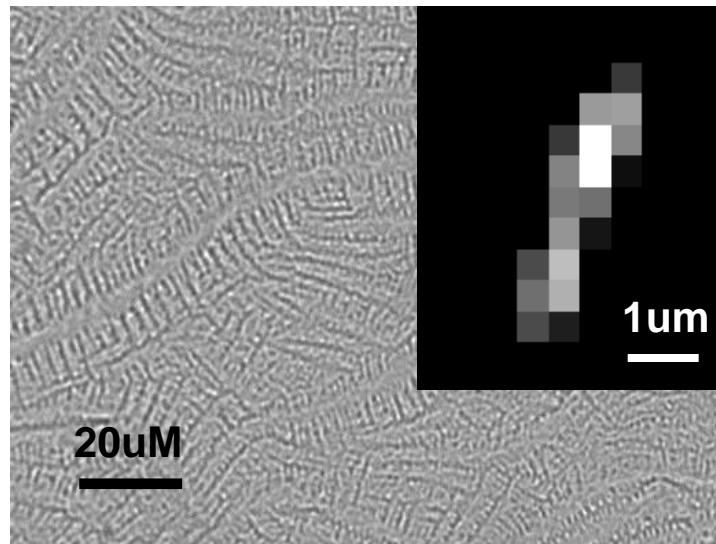
Single Molecule Detection In-Vivo using a Collagen Platform



Hong
Jin



SWNT embedded into type 1 collagen film



Red line: fit to
Hidden Markov
model

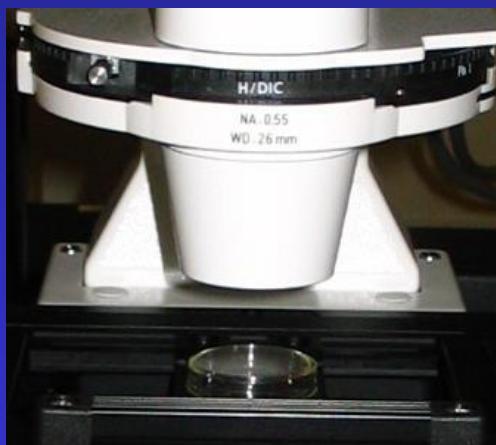
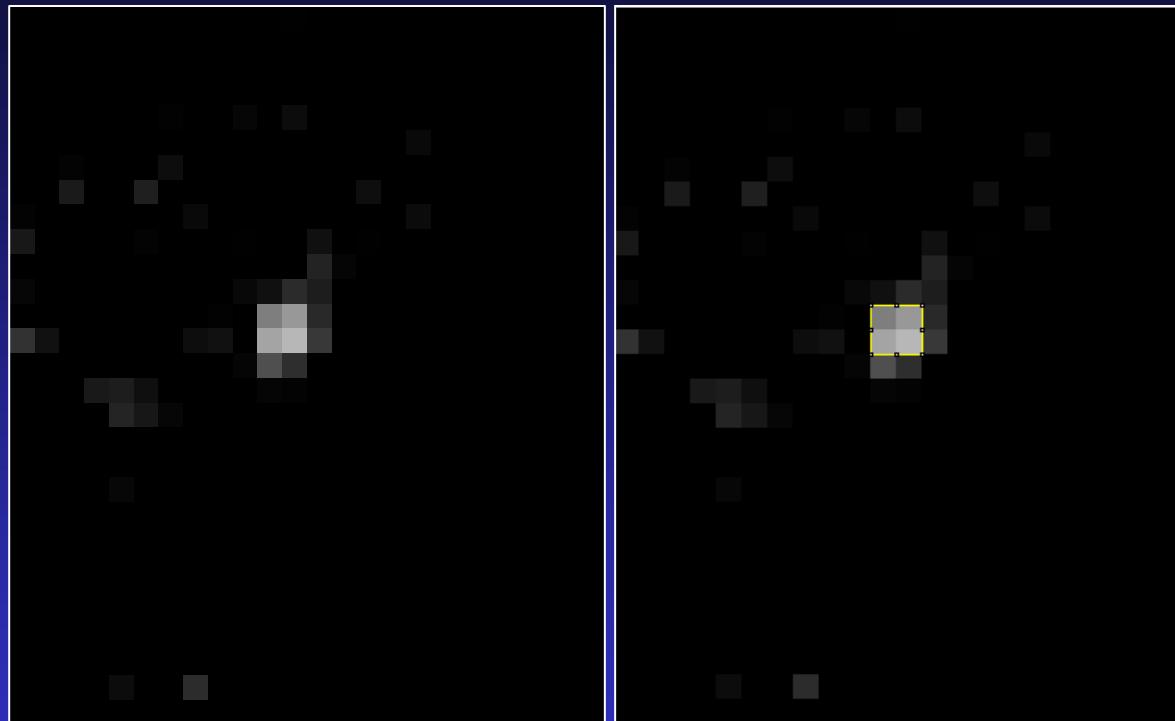
McKinney, S. A.; Joo, C.; Ha,
T. *Biophysical Journal*
2006, 91, 1941-1951.

Single Molecule Detection of Hydrogen Peroxide

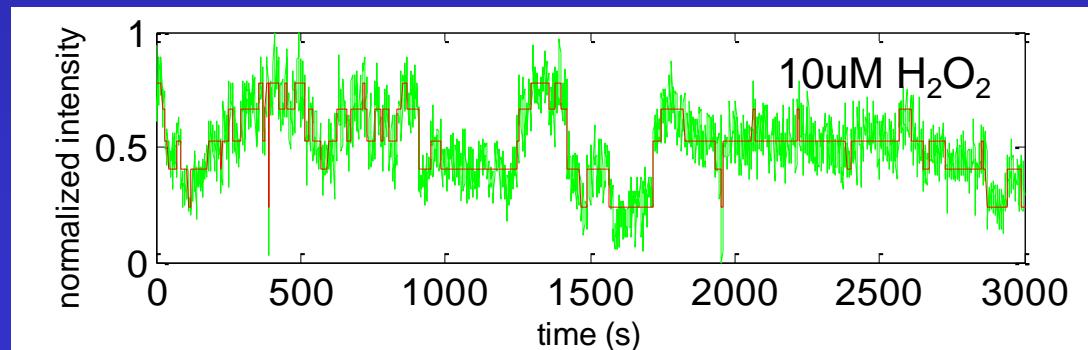
InGaAs nIR Detector
(0.7 to 1.9 μm)



Single isolated SWNT in collagen film, 10 $\mu\text{M} \text{H}_2\text{O}_2$

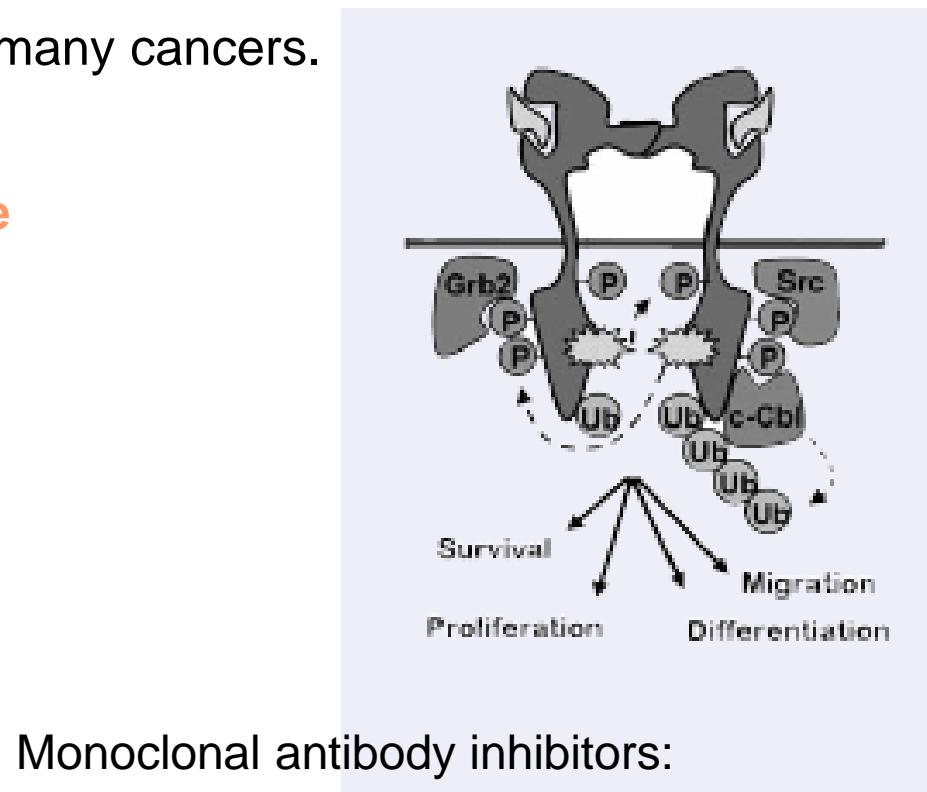
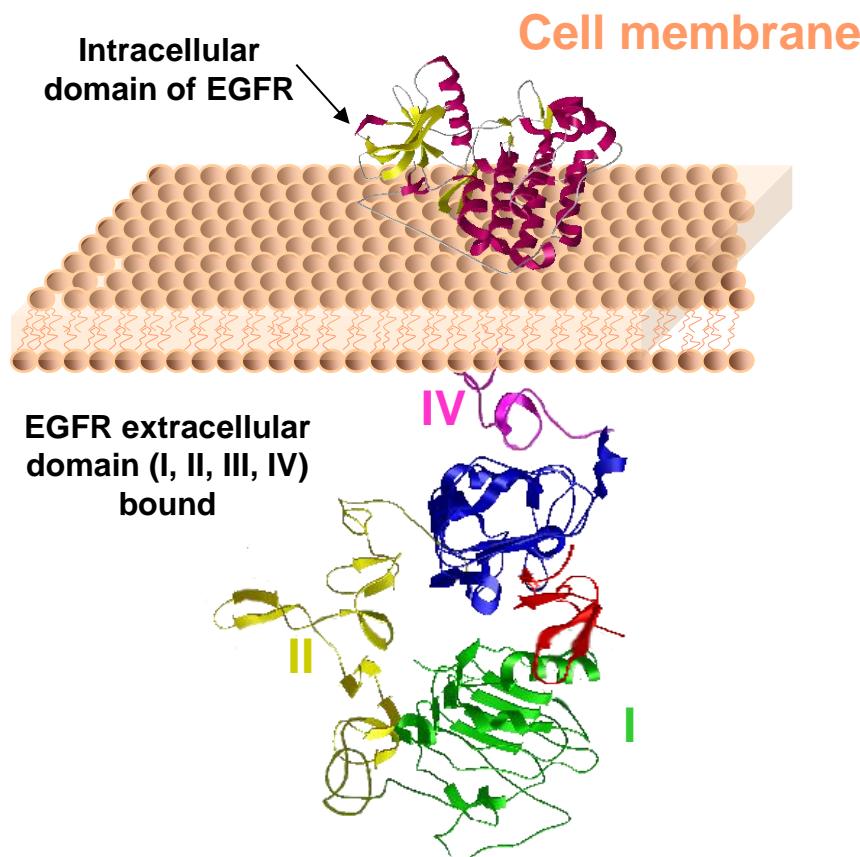


Objective:
1.46 NA (TIR)



Biological Problem: Epidermal Growth Factor Receptor

Epidermal Growth Factor Receptor (EGF): extracellular membrane protein; tyrosine kinase; ErbB family of receptors; modulates cell proliferation
EGFR over-expression implicated in many cancers.



Monoclonal antibody inhibitors:

Cetuximab and Panitumumab

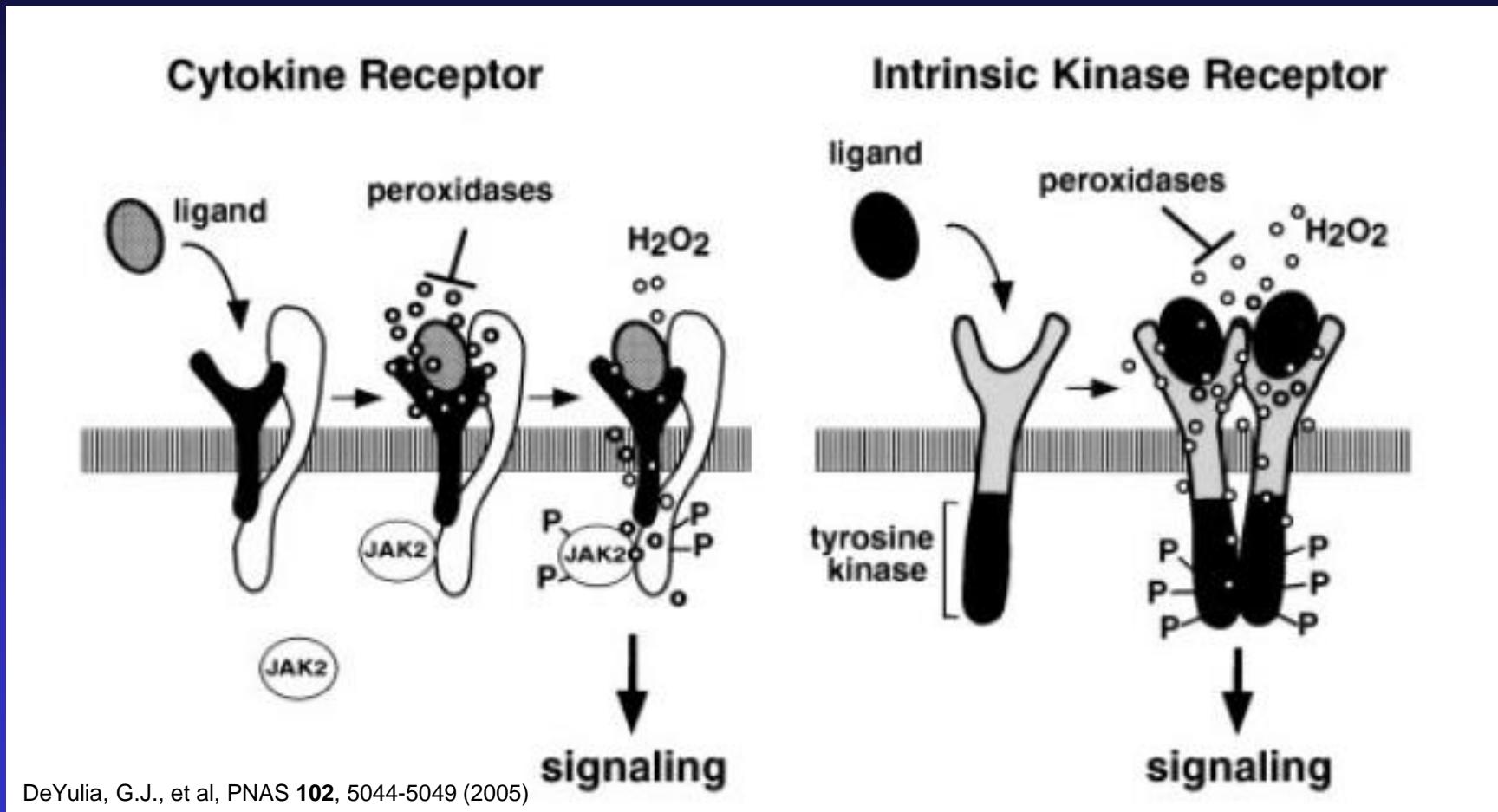
> \$620 million/year

Questions about H₂O₂ signaling pathway in EGFR

Source of H₂O₂ generation in response to EGF stimulation

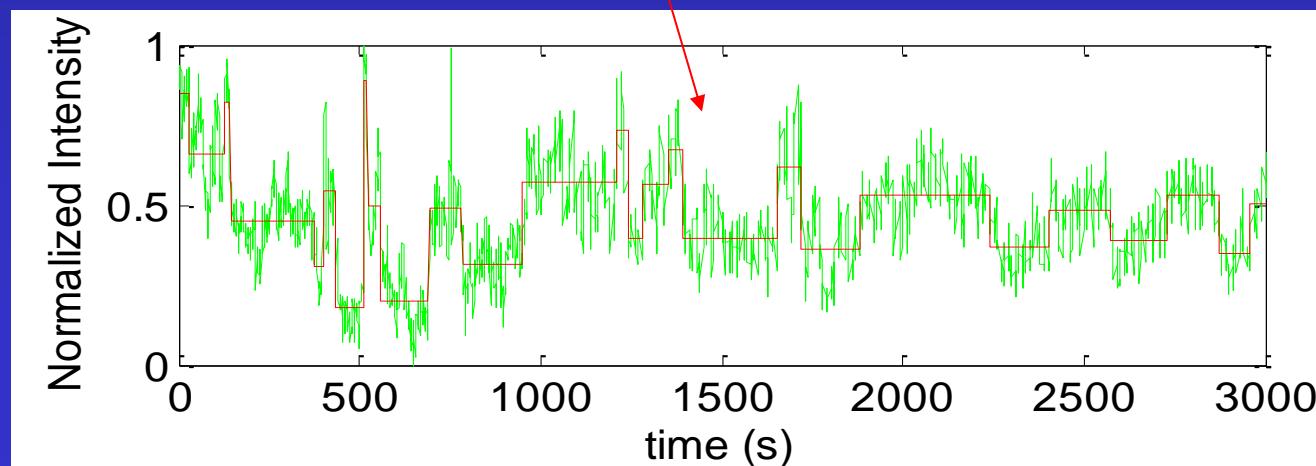
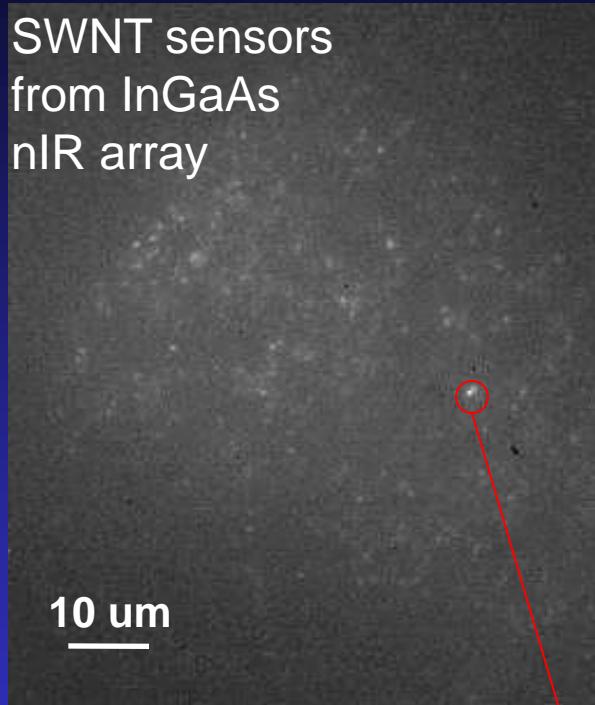
Location within the cell

Relationship to better known phosphorylation pathway



Goal of our work: develop a platform to study ROS signaling at the single cell, single molecule level

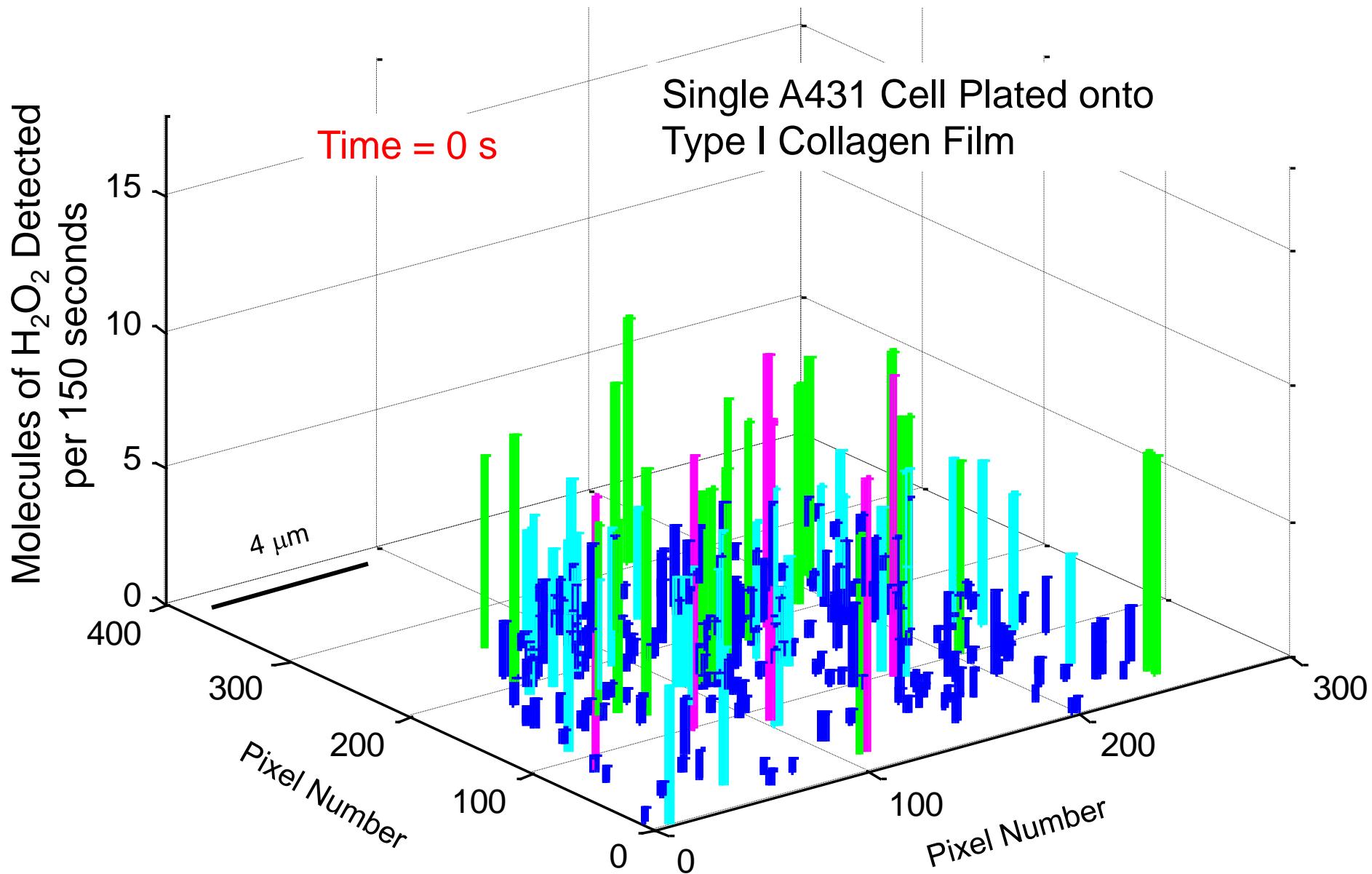
Spatial and Temporal Single Molecule Map from Live A431 Human Epidermoid Carcinoma



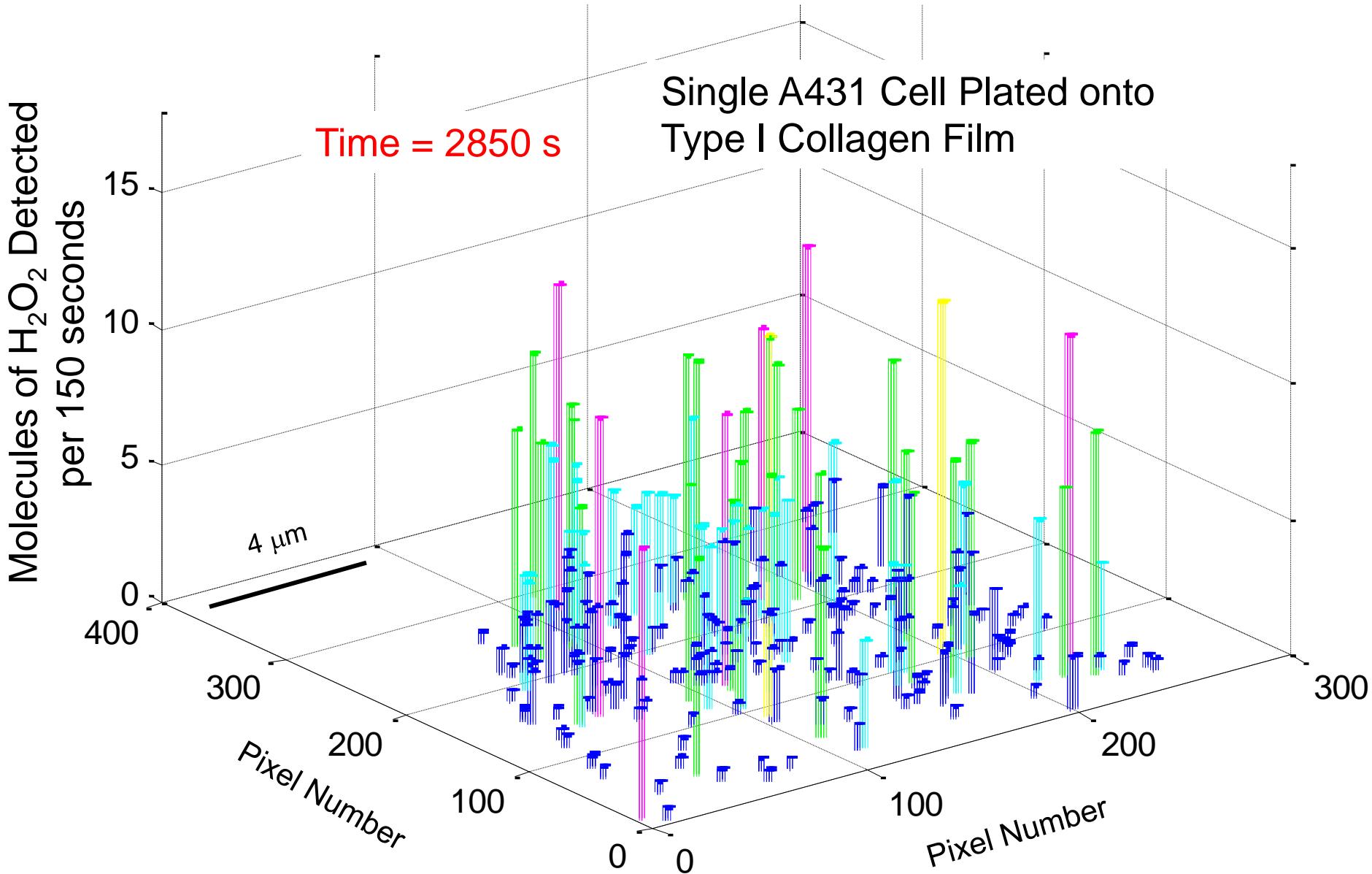
One example trace, from a 4 pixel region of interest (ROI) on one single nanotube.

The goal is to examine all the ROIs and map the spatial behavior in real time.

Real Time, Spatially Resolved Single Molecule Detection of H₂O₂ from a single A431 Cell

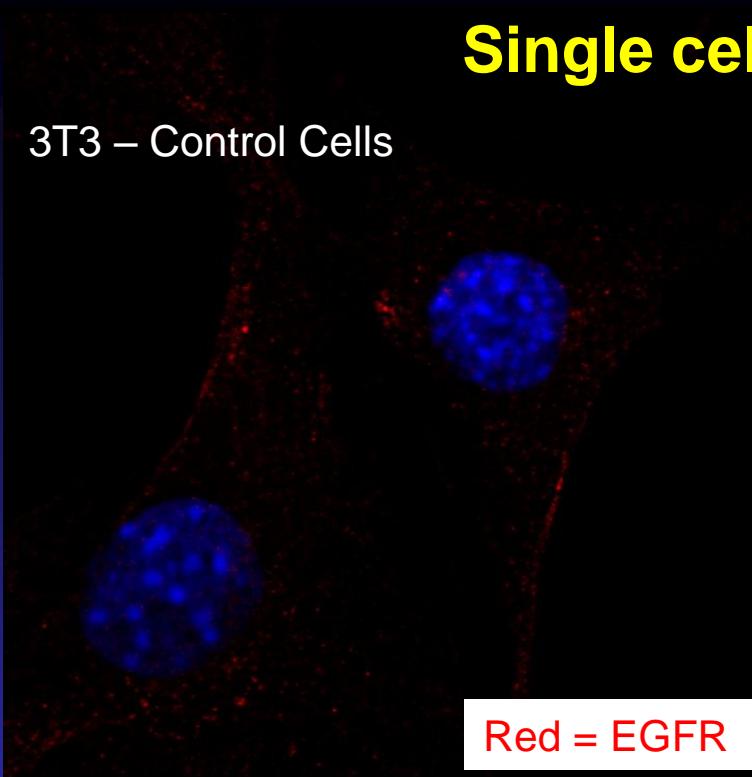


Real Time, Spatially Resolved Single Molecule Detection of H₂O₂ from a single A431 Cell



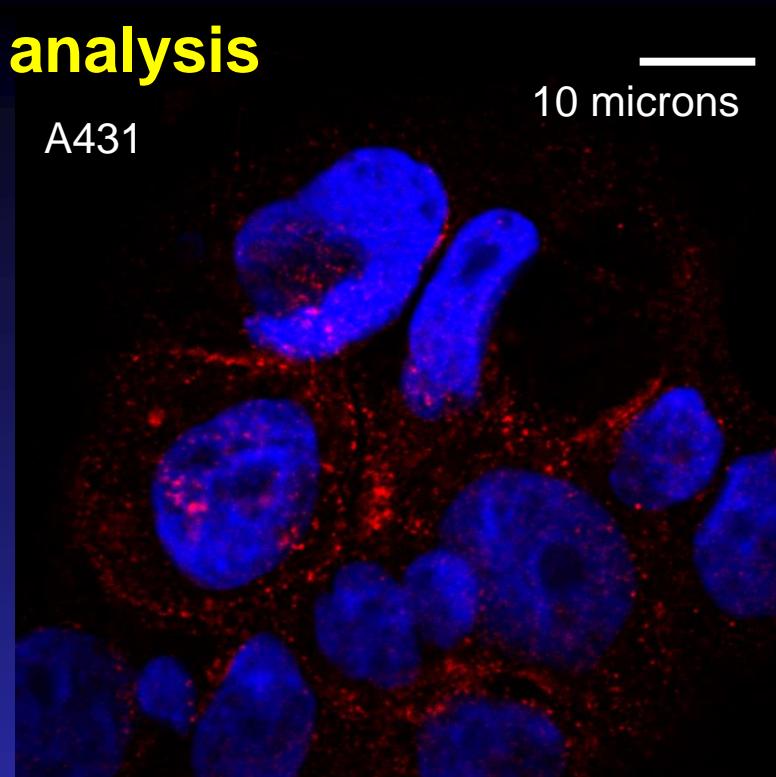
Single cell analysis

3T3 – Control Cells

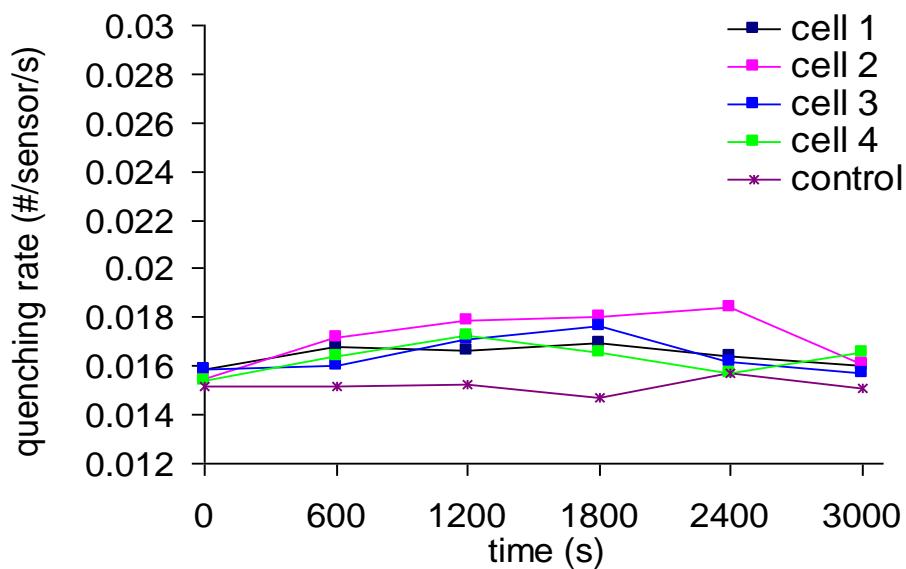


Red = EGFR

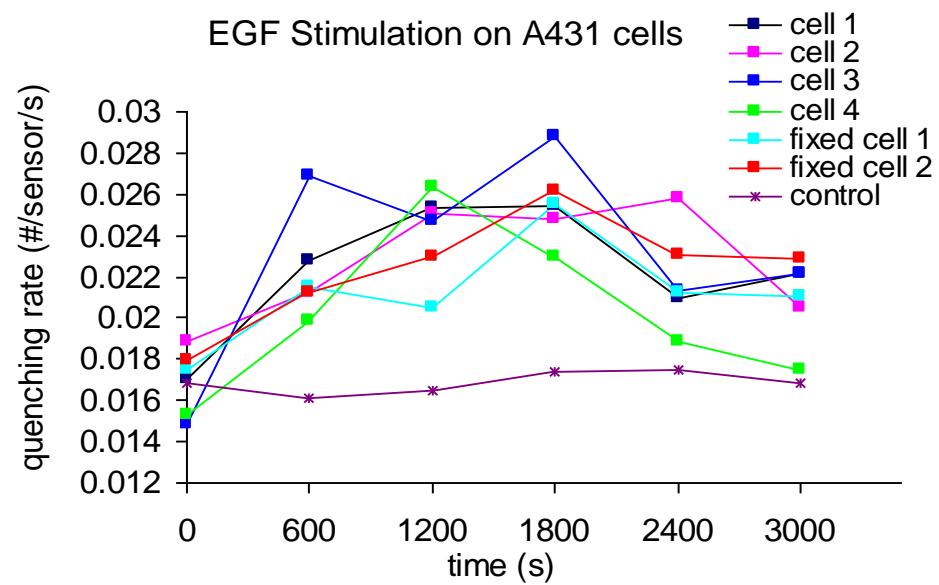
A431



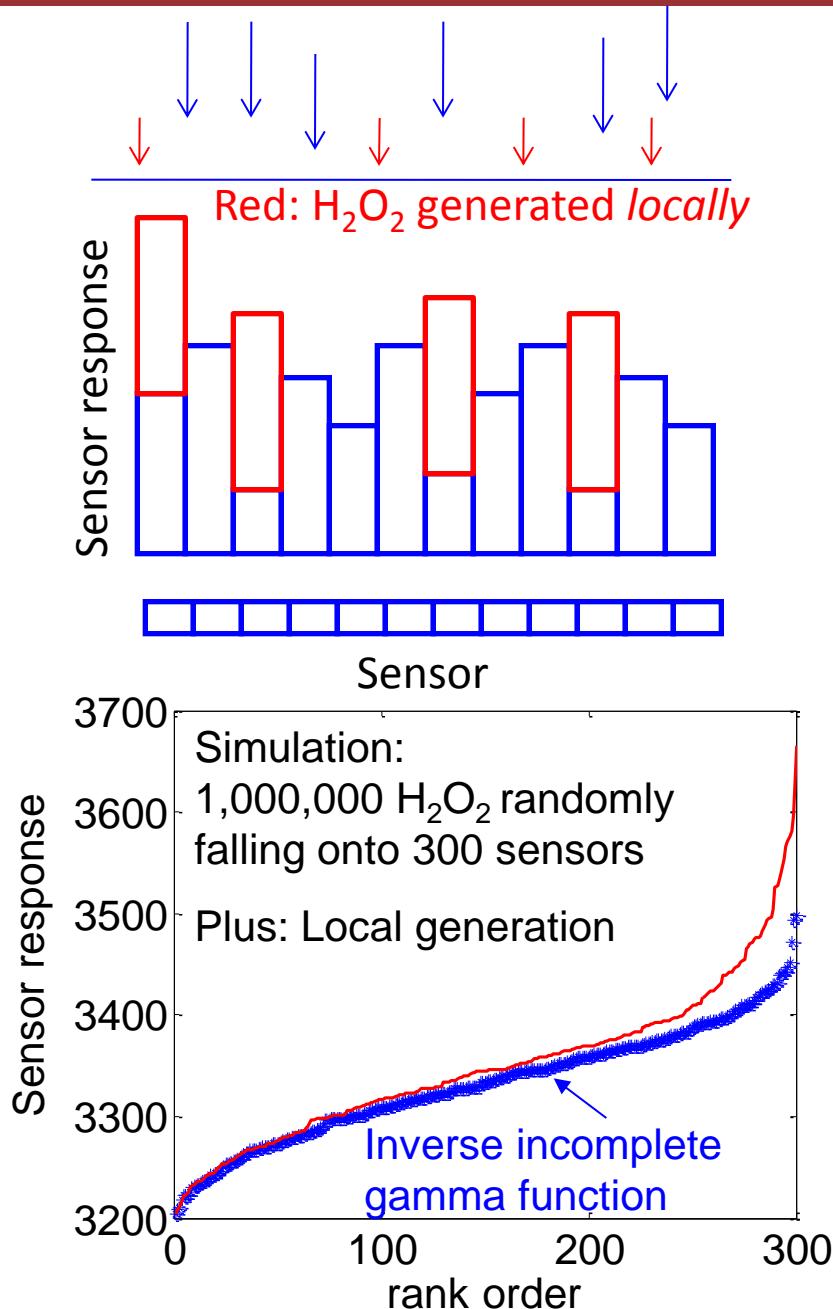
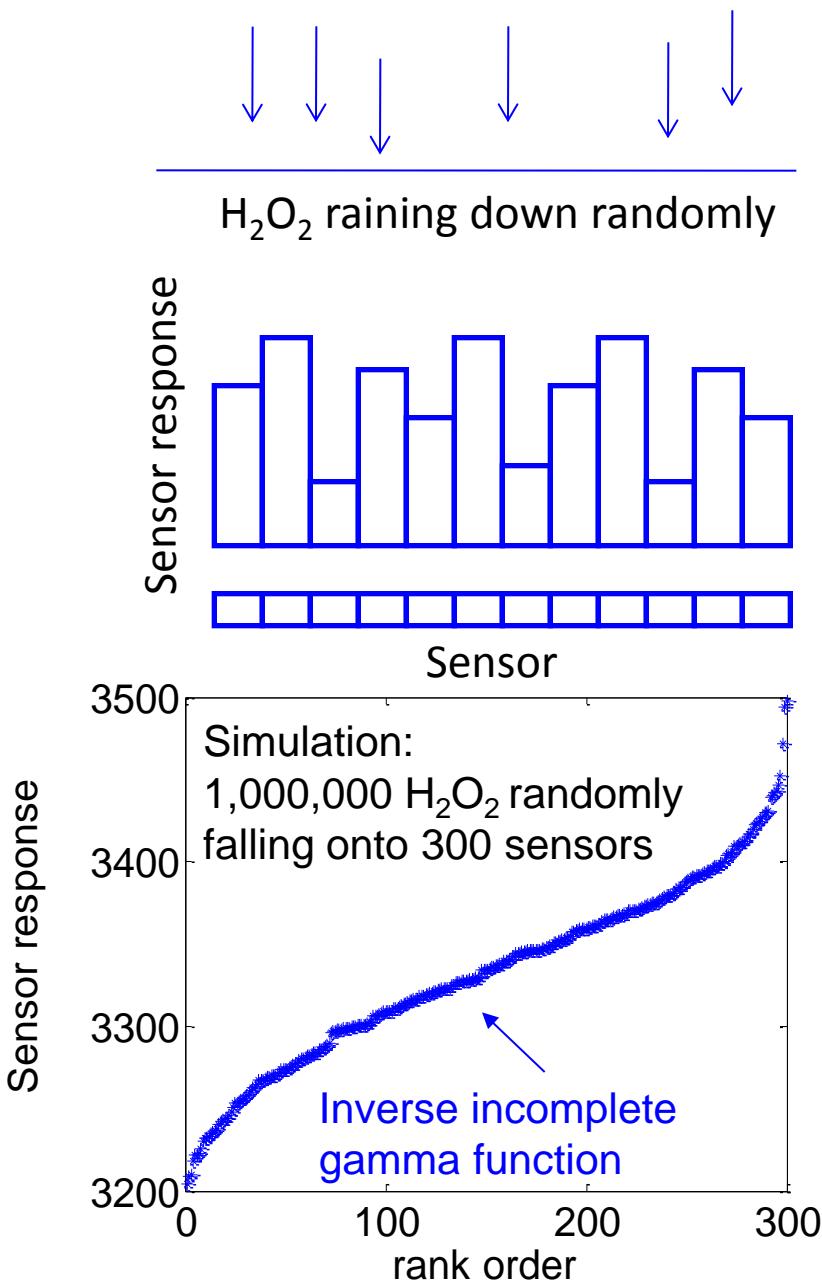
EGF Stimulation on 3T3 cells



EGF Stimulation on A431 cells



Sensors can distinguish background/membrane generation!



Counting molecules originating from the membrane

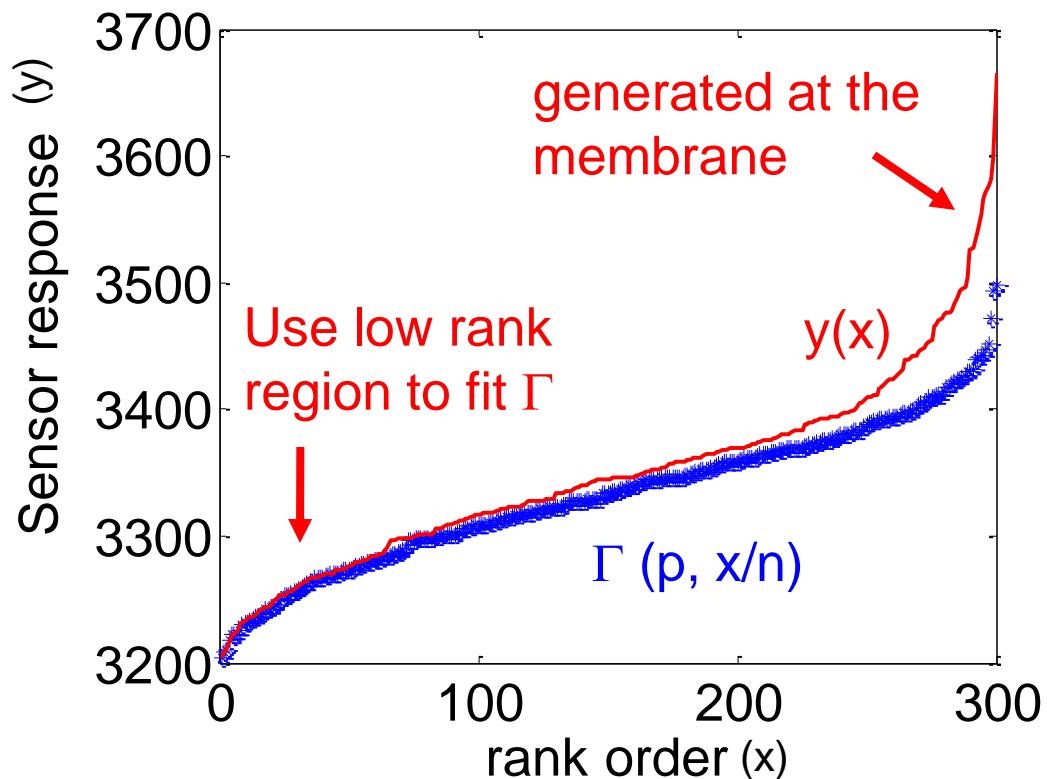
Given any distribution of sensor responses (rank ordered), $y(x)$, the curve at $x \rightarrow 0$ specifies the binomial background (gamma function):

$$y_{membrane}(x) = y(x) - \Gamma\left(p, \frac{x}{n}\right)$$

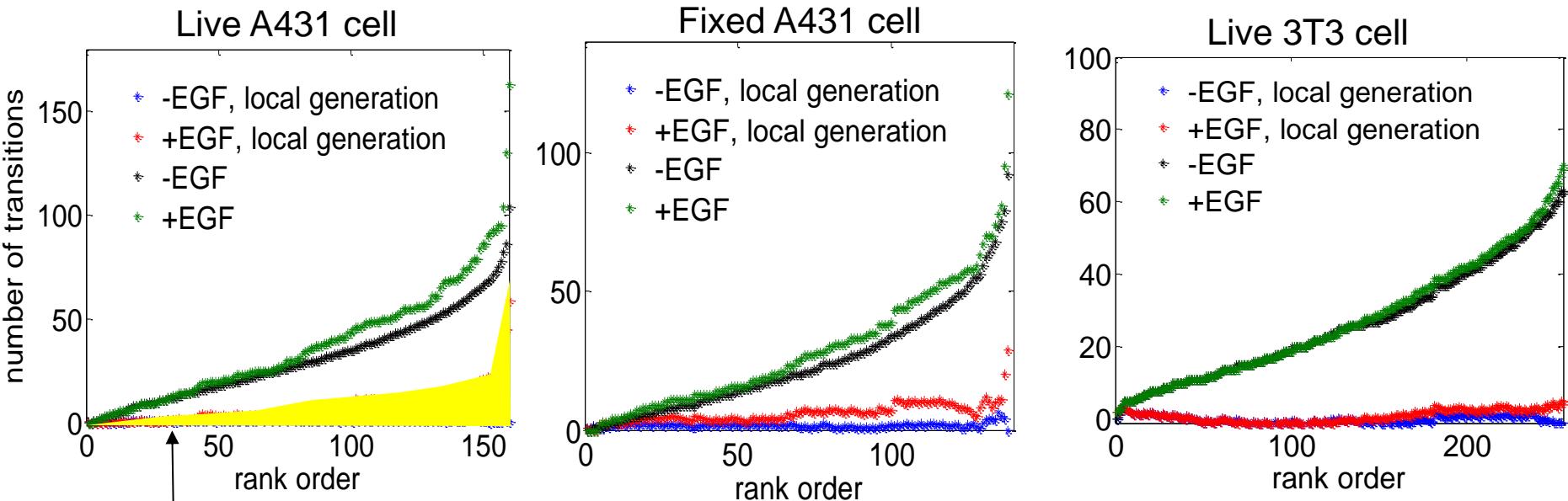
Γ : incomplete gamma function
(background)

p : mean value of background distribution, obtained from fitting the small number of data points at $x \rightarrow 0$

n = number of sensors



Separating EGFR Effect From Background H₂O₂



Activity increase arises from EGF-EGFR binding

Similar between live and fixed A431 cells

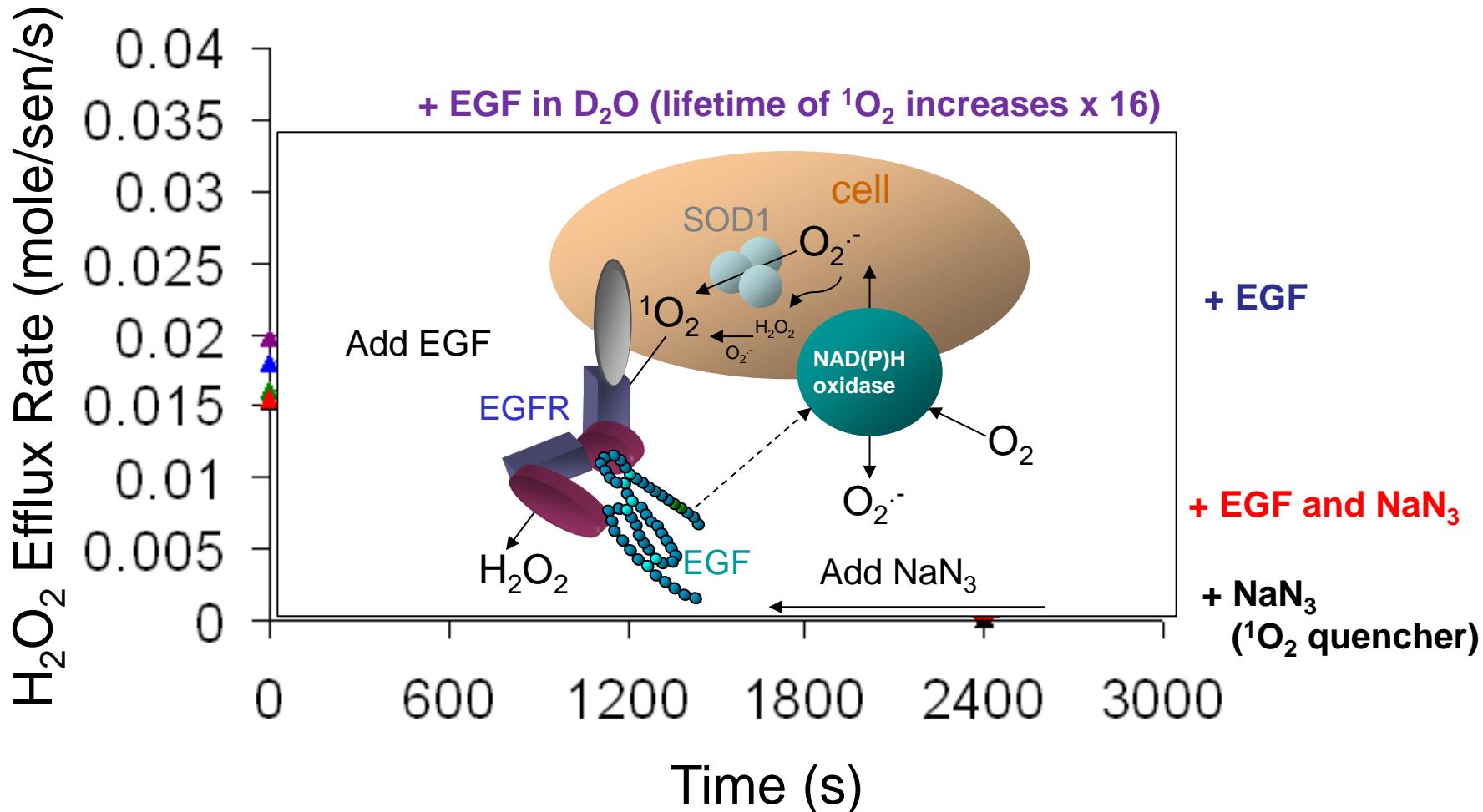
Increase in activity is more evident in A431 than 3T3

Receptor activity:

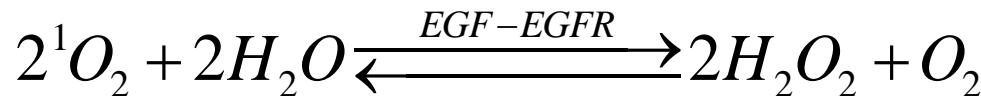
SWNT arrays measure receptor activity in real time on a single cell!



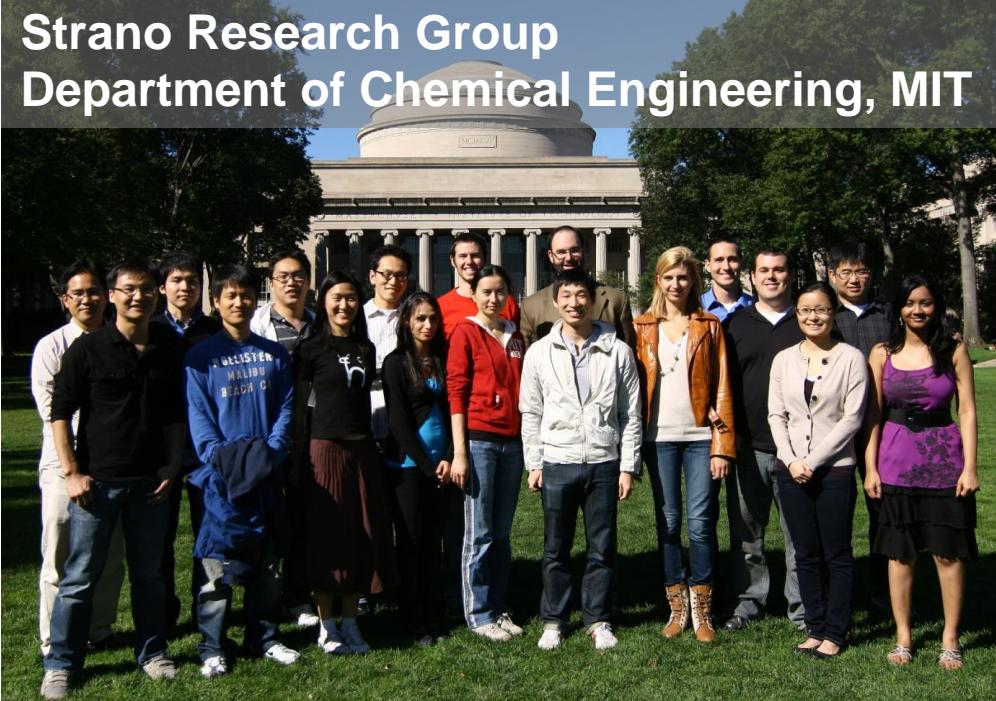
Singlet Oxygen (${}^1\text{O}_2$) Oxidation is the H_2O_2 Source



Generation on
receptor



Acknowledgements



Collaborators

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Prof. Tomas Palacios (EECS, MIT)
Prof. Mildred Dresselhaus (Physics, EECS, MIT)
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Akira Kudo
Sayalee Mahajan
Geraldine Paulus
Nigel Reuel
Chih-Jen Shih
Steven Shimizu
Zachary Ulissi
Jingqing Zhang